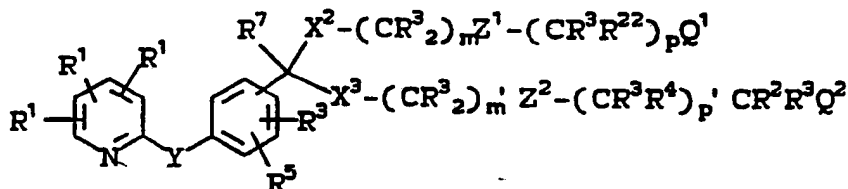




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<p>(21) International Application Number: PCT/CA93/00145</p> <p>(22) International Filing Date: 2 April 1993 (02.04.93)</p> <p>(30) Priority data: 866,697 13 April 1992 (13.04.92) US</p> <p>(71) Applicant: MERCK FROSST CANADA INC. [CA/CA]; 16711 Trans-Canada Highway, Kirkland, Quebec H9H 3L1 (CA).</p> <p>(72) Inventors: ZAMBONI, Robert ; 114 Champ d'Avion, Point-Claire, Quebec H9R 5S7 (CA). GUAY, Daniel ; 67 Roger Maillet, Notre Dame de Ile Perrot, Quebec J7V 8P2 (CA). GAUTHIER, Jacques-Yves ; 540 Odette Oigny, Apt. 2, Laval, Quebec H7N 5Z4 (CA).</p>		<p>(74) Agent: MURPHY, Kevin, P.; Suite 800, 1001 de Maison- neuve Blvd. West, Montreal, Quebec H3A 3C8 (CA).</p> <p>(81) Designated States: AU, BB, BG, BR, CA, CZ, FI, HU, JP, KR, LK, MG, MN, MW, NO, NZ, PL, RO, RU, SD, SK, UA, European patent (AT, BE, CH, DE, DK, ES, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE), OAPI pa- tent (BF, BJ, CF, CG, CI, CM, GA, GN, ML, MR, NE, SN, TD, TG).</p> <p>Published <i>With international search report.</i></p>

(54) Title: PYRIDINE-SUBSTITUTED BENZYL ALCOHOLS AS LEUKOTRIENE ANTAGONISTS



(I)

## (57) Abstract

Compounds having formula (I) are antagonists of the actions of leukotrienes. These compounds are useful as anti-asthmatic, anti-allergic, anti-inflammatory, and cytoprotective agents. They are also useful in treating angina, cerebral spasm, glomerular nephritis, hepatitis, endotoxemia, uveitis, and allograft rejection.

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PYRIDINE-SUBSTITUTED BENZYL ALCOHOLS AS  
LEUKOTRIENE ANTAGONISTS

BACKGROUND OF THE INVENTION

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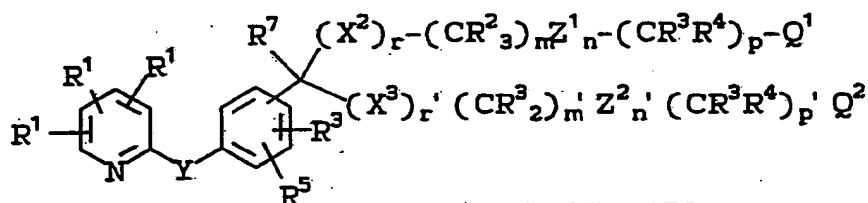
30

The leukotrienes constitute a group of locally acting hormones, produced in living systems from arachidonic acid. The major leukotrienes are Leukotriene B<sub>4</sub> (abbreviated as LTB<sub>4</sub>), LTC<sub>4</sub>, LTD<sub>4</sub>, and LTE<sub>4</sub>. The biosynthesis of these leukotrienes begins with the action of the enzyme 5-lipoxygenase on arachidonic acid to produce the epoxide known as Leukotriene A<sub>4</sub> (LTA<sub>4</sub>), which is converted to the other leukotrienes by subsequent enzymatic steps. Further details of the biosynthesis, as well as the metabolism of the leukotrienes, are to be found in the book Leukotrienes and Lipoxygenases, ed. J. Rokach, Elsevier, Amsterdam (1989). The actions of the leukotrienes in living systems and their contribution to various diseases states are also discussed in the book by Rokach.

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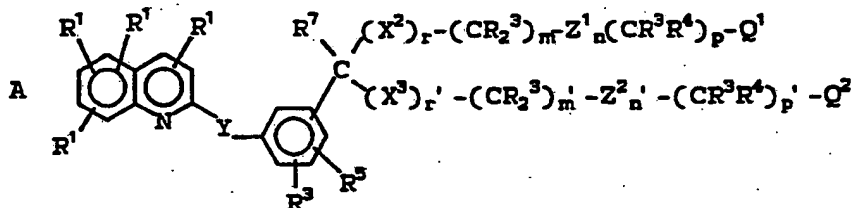
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U.S. Patent 5,004,743 discloses structures of leukotriene antagonists which differ from the present compounds, most notably in the absence of the benzyl alcohol and of fused cycloalkyl pyridines. The structure of the compounds disclosed in the above patent application is shown below.

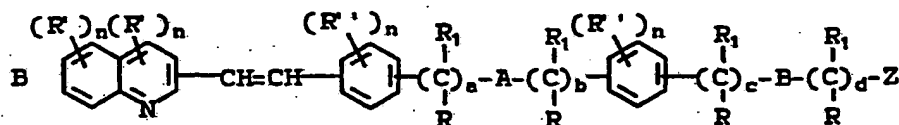


U. S. P. 5, 004, 473

The art also describes certain quinoline-containing compounds as having leukotriene antagonist activity. Thus, EP 318,093 (Merck) describes compounds of structure A while compounds of structure B are disclosed in WO 89/12629 (Rorer).



EP 318, 093 (Merck)



WO 89/12629 (Rorer)

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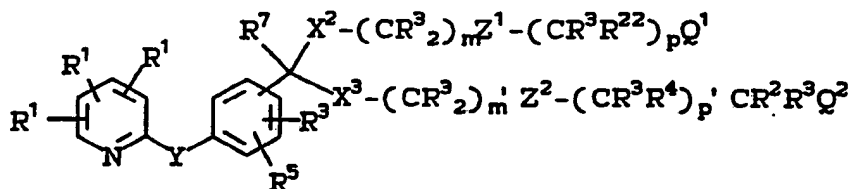
SUMMARY OF THE INVENTION

The present invention relates to pyridine-substituted benzyl alcohols having activity as leukotriene antagonists, to methods for their preparation, and to methods and pharmaceutical formulations for using these compounds in mammals (especially humans).

Because of their activity as leukotriene antagonists, the compounds of the present invention are useful as anti-asthmatic, anti-allergic, anti-inflammatory, and cytoprotective agents. They are also useful in treating angina, cerebral spasm, glomerular nephritis, hepatitis, endotoxemia, uveitis, and allograft rejection.

DETAILED DESCRIPTION OF THE INVENTION

The compounds of this invention are best realized by Formula I:



I

wherein:

$R^1$  is H, halogen, CN, lower alkyl, cyloalkyl, polyhalo lower alkyl, lower alkoxy, lower alkoxy lower alkyl, lower alkylthio lower alkyl, lower alkenyl, substituted or unsubstituted phenyl, pyridyl, thiazolyl,

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oxazoly1, furanyl or thienyl, or adjacent  $R^1$ 's and the carbons through which they are attached may form a saturated ring of 5 to 10 carbon atoms;

- 5  ~~$R^2$  is~~ lower alkyl, lower alkenyl, lower alkynyl,  $-CF_3$ ,  $-CH_2F$ ,  $-CHF_2$ ,  $-CH_2CF_3$ , substituted or unsubstituted phenyl, substituted or unsubstituted benzyl, substituted or unsubstituted 2-phenethyl, or two  $R^2$  groups joined to the same carbon may form a saturated ring of up to 8 members containing 0 to 2 heteroatoms chosen from O, S, and N;
- 10  $R^3$  is H or  $R^2$ ;  
 $CR^3R^{22}$  may be the radical of a standard amino acid;
- 15  $R^4$  is halogen,  $-NO_2$ ,  $-CN$ ,  $-OR^3$ ,  $-SR^3$ ,  $NR^3R^3$ ,  $NR^3C(O)R^7$ , or  $R^3$ ;
- $R^5$  is H, halogen,  $-NO_2$ ,  $-N_3$ ,  $-CN$ ,  $-SR^2$ ,  $-NR^3R^3$ ,  $-OR^3$ , lower alkyl, or  $-C(O)R^3$ ;
- 20  $R^6$  is  $-(CH_2)_s-C(R^7R^7)-(CH_2)_s-R^8$  or  $-CH_2C(O)NR^{12}R^{12}$ ;
- $R^7$  is H or lower alkyl;
- $R^8$  is A) a monocyclic or bicyclic heterocyclic radical containing from 3 to 12 nuclear carbon atoms and 1 or 2 nuclear heteroatoms selected from N, S or O and with each ring in the heterocyclic radical being formed of 5 or 6 atoms, or
- 25 B) the radical  $W-R^9$ ;
- $R^9$  contains up to 20 carbon atoms and is (1) an alkyl group or (2) an alkylcarbonyl group of an organic acyclic or monocyclic carboxylic acid containing not more than 1 heteroatom in the ring;
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- $R^{10}$  is  $-SR^{11}$ ,  $-OR^{12}$ , or  $-NR^{12}R^{12}$ ;  
 $R^{11}$  is lower alkyl,  $-C(O)R^{14}$ , unsubstituted phenyl,  
or unsubstituted benzyl;  
 $R^{12}$  is H,  $R^{11}$ , or two  $R^{12}$  groups joined to the same  
5 N may form a saturated ring of 5 or 6  
members containing up to two heteroatoms  
chosen from O, S, and N;  
 $R^{13}$  is lower alkyl, lower alkenyl, lower alkynyl,  
10  $-CF_3$ , or substituted or unsubstituted  
phenyl, benzyl, or 2-phenethyl;  
 $R^{14}$  is H or  $R^{13}$ ;  
 $R^{15}$  is  $R^3$  or halogen;  
 $R^{16}$  is H, lower alkyl, or OH;  
 $R^{17}$  is lower alkyl, lower alkenyl, lower alkynyl,  
15 or substituted or unsubstituted phenyl,  
benzyl, or 2-phenethyl;  
 $R^{18}$  is lower alkyl, lower alkenyl, lower alkynyl,  
 $-CF_3$ , or substituted or unsubstituted  
phenyl, benzyl, or 2-phenethyl;  
20  $R^{19}$  is lower alkyl, lower alkenyl, lower alkynyl,  
 $-CF_3$ , or substituted or unsubstituted  
phenyl, benzyl, or 2-phenethyl;  
 $R^{20}$  is H, lower alkyl, substituted or unsubstituted  
phenyl, benzyl, phenethyl, or pyridinyl, or  
25 two  $R^{20}$  groups joined to the same N may form  
a saturated ring of 5 or 6 members  
containing one to two heteroatoms chosen  
from O, S, and N;  
 $R^{21}$  is H or  $R^{17}$ ;  
30  $R^{22}$  is  $R^4$ ,  $CHR^7OR^3$ , or  $CHR^7SR^2$ ;  
m and m' are independently 0-8;  
p and p' are independently 0-8;  
m + p is 1-10 when  $X^2$  is O, S,  $S(O)$ , or  $S(O)_2$ ;  
m + p is 0-10 when  $X^2$  is  $CR^3R^{16}$  or a bond;

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$m' + p'$  is 0-10;

$s$  is 0-3;

$Q^1$  is  $-C(O)OR^3$ , 1H (or 2H)-tetrazol-5-yl,  
 $-C(O)OR^6$ ,  $-C(O)NHS(O)_2R^{13}$ ,  $-CN$ ,  
 $-C(O)NR^{12}R^{12}$ ,  $NR^{21}S(O)_2R^{13}$ ,  
 $-NR^{12}C(O)NR^{12}R^{12}$ ,  $-NR^{21}C(O)R^{18}$ ,  
 $OC(O)NR^{12}R^{12}$ ,  $-C(O)R^{19}$ ,  $-S(O)R^{18}$ ,  $-S(O)_2R^{18}$ ,  
 $-S(O)_2NR^{12}R^{12}$ ,  $-NO_2$ ,  $NR^{21}C(O)OR^{17}$ ,  
 $-C(NR^{12}R^{12})=NR^{12}$ , or  $-C(R^{13})=NOH$ ; or if  $Q^1$   
 is  $C(O)OH$  and  $R^{22}$  is  $-OH$ ,  $-SH$ ,  $CHR^7OH$  or  
 $-NHR^3$ , then  $Q^1$  and  $R^{22}$  and the carbons  
 through which they are attached may form a  
 heterocyclic ring by loss of water;

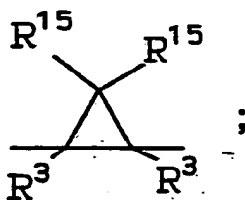
$Q^2$  is  $OR^3$ ;

$W$  is  $O$ ,  $S$ , or  $NR^3$ ;

$X^1$  is  $O$ ,  $S$ ,  $-S(O)-$ ,  $-S(O)_2-$ ,  $-N(R^3)-$ , or  
 $-CR^3R^3-$ ;

$X^2$  and  $X^3$  are independently  $O$ ,  $S$ ,  $S(O)$ ,  $S(O)_2$ ,  
 $CR^3R^{16}$ , or a bond;

$Y$  is  $-CR^3=CR^3-$ ,  $-C\equiv C-$ ,  $-CR^3R^3-X^1-$ ,  $-X^1-CR^3R^3-$ ,  
 $-CR^3R^3-X^1-CR^3R^3-$ ,  
 $-C(O)-$ ,  $-NR^3C(O)-$ ,  $-C(O)NR^3-$ ,  $O$ ,  $S$ ,  $NR^3$ ,  
 or



$Z^1$  and  $Z^2$  are independently  $-HET(-R^3-R^5)-$  or a bond;

$HET$  is the diradical of a benzene, a pyridine, a  
 furan, or a thiophene;

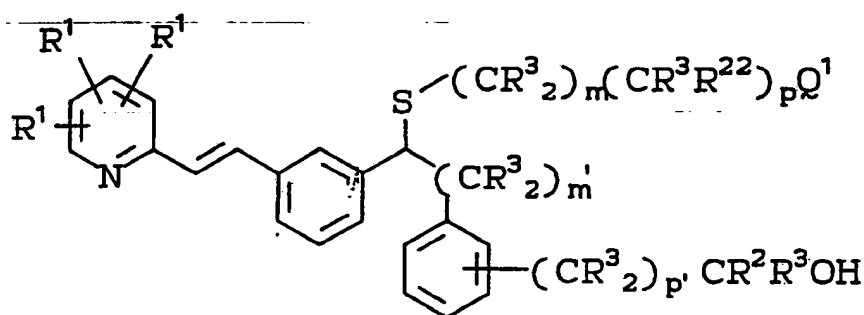
or a pharmaceutically acceptable salt thereof.

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More preferred compounds of Formula I are represented by Formula Ia:



Ia

wherein:

$R^1$  is H, halogen, lower alkyl, polyhalo lower alkyl, lower alkoxy or adjacent  $R^1$ 's and the carbons through which they are attached may form a saturated ring of 5 to 7 carbon atoms;

$R^{22}$  is  $R^3$ ,  $-\text{CH}_2\text{OR}^3$ , or  $-\text{CH}_2\text{SR}^2$ ;

$Q^1$  is  $-\text{C}(\text{O})\text{OH}$ , 1H(or 2H)-tetrazol-5-yl,  $-\text{C}(\text{O})\text{NHS}(\text{O})_2\text{R}^{13}$ ,  $-\text{C}(\text{O})\text{NR}^{12}\text{R}^{12}$ , or  $-\text{NHS}(\text{O})_2\text{R}^{13}$ ;

$m'$  is 2 or 3;

$p'$  is 0 or 1;

$m + p$  is 1-5; and

the remaining definitions are as in Formula I;

or a pharmaceutically acceptable salt thereof.

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The following abbreviations have the indicated meanings:

AIBN = 2,2'-azobis(isobutyronitrile)  
5 Py = 2-, 3-, or 4-pyridyl  
Fu = 2- or 3-furanyl  
Et = ethyl  
Me = methyl  
Bz = benzyl  
10 Ph = phenyl  
t-Bu = tert-butyl  
i-Pr = isopropyl  
n-Pr = normal propyl  
c-Hex = cyclohexyl  
15 c-Pr = cyclopropyl  
c- = cyclo  
Ac = acetyl  
Tz = tetrazol-5-yl  
Th = 2- or 3-thienyl  
20 C<sub>3</sub>H<sub>5</sub> = allyl  
i-C<sub>3</sub>H<sub>5</sub> = 2-propenyl  
c-Pen = cyclopentyl  
c-Bu = cyclobutyl  
PPTS = pyridinium p-toluene sulfonate  
25 phe = benzenediyl  
NBS = N-bromosuccinimide  
NCS = N-chlorosuccinimide  
pye = pyridinediyl  
PTSA = p-toluenesulfonic acid  
30 Thia = thiazolyl  
Ox = oxazolyl  
fur = furandiyl

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r.t. = room temperature  
thio = thiophenediyl  
DHP = 4H-2,3-dihdropyran  
THP = tetrahydropyran

5

The terms alkyl, alkenyl, and alkynyl mean linear and branched structures and combinations thereof.

10 The term "alkyl" includes "lower alkyl" and extends to cover carbon fragments having up to 20 carbon atoms. Examples of alkyl groups include octyl, nonyl, undecyl, dodecyl, tridecyl, tetradecyl, pentadecyl, eicosyl, 3,7-diethyl-2,2-dimethyl-4-propylnonyl, and the like.

15 The term "polyhalo" means one or more hydrogen atoms are replaced by halogen atoms.

The term "lower alkyl" means alkyl groups of from 1 to 7 carbon atoms. Examples of lower alkyl groups include methyl, ethyl, propyl, isopropyl, 20 butyl, s- and t-butyl, pentyl, hexyl, heptyl, and the like.

The term "polyhalo lower alkyl" means a lower alkyl group in which one or more of the hydrogen atoms has been replaced by a halogen atom.

25 The term "cycloalkyl" refers to a hydrocarbon, containing one or more rings of from 3 to 12 carbon atoms, with the hydrocarbon having up to a total of 20 carbon atoms. Examples of cycloalkyl groups are cyclopropyl, cyclopentyl, cycloheptyl, 30 aldamantyl, cyclododecylmethyl, 2-ethyl-1-bicyclo[4.4.0]decyl, and the like.

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The term "alkenyl" includes "lower alkenyl" and means alkenyl groups of 2 to 20 carbon atoms. Examples of alkenyl groups include allyl, 5-decen-1-yl, 2-dodecen-1-yl, and the like.

5 "Lower alkenyl" means alkenyl groups of 2 to 7 carbon atoms. Examples of lower alkenyl groups include vinyl, allyl, isopropenyl, pentenyl, hexenyl, heptenyl, 1-propenyl, 2-butenyl, 2-methyl-2-butenyl, and the like.

10 "Cycloalkenyl" means alkenyl groups of 3 to 20 carbon atoms, which include a ring of 3 to 12 carbon atoms, and in which the alkenyl double bond may be located anywhere in the structure. Examples of cycloalkenyl groups are cyclopropen-1-yl, 15 cyclohexen-3-yl, 2-vinyladamant-1-yl, 5-methylenedodec-1-yl, and the like.

The term "alkynyl" includes "lower alkynyl" and means alkynyl groups of 2 to 20 carbon atoms. Examples of alkynyl groups are ethynyl, 20 2-pentadecyn-1-yl, 1-eicosyn-1-yl, and the like.

"Lower alkynyl" means alkynyl groups of 2 to 7 carbon atoms. Examples of lower alkynyl groups include ethynyl, propargyl, 3-methyl-1-pentynyl, 2-heptynyl, and the like.

25 The term "cycloalkynyl" means alkynyl groups of 5 to 20 carbon atoms, which include a ring of 3 to 20 carbon atoms. The alkynyl triple bond may be located anywhere in the group, with the proviso that if it is within a ring, such a ring must be of 10 30 members or greater. Examples of cycloalkynyl are cyclododecyn-3-yl, 3-cyclohexyl-1-propyn-1-yl, and the like.

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6           The term "lower alkoxy" means alkoxy groups  
of from 1 to 7 carbon atoms of a straight, branched,  
or cyclic configuration. Examples of lower alkoxy  
groups include methoxy, ethoxy, propoxy, isopropoxy,  
5   cyclopropyloxy, cyclohexyloxy, and the like.

          The term "lower alkylthio" means alkylthio  
groups of from 1 to 7 carbon atoms of a straight,  
branched or cyclic configuration. Examples of lower  
alkylthio groups include methylthio, propylthio,  
10   isopropylthio, cycloheptylthio, etc. By way of  
illustration, the propylthio group signifies  
-SCH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>.

          The term "lower alkylsulfonyl" means  
alkylsulfonyl groups of from 1 to 7 carbon atoms of a  
15   straight, branched, or cyclic configuration.  
Examples of lower alkylsulfonyl groups are methyl-  
sulfonyl, 2-butylsulfonyl, cyclohexylmethylsulfonyl,  
etc. By way of illustration, the 2-butylsulfonyl  
group signifies -S(O)<sub>2</sub>CH(CH<sub>3</sub>)CH<sub>2</sub>CH<sub>3</sub>.

20           "Alkylcarbonyl" includes "lower  
alkylcarbonyl" and means alkylcarbonyl groups of 1 to  
20 carbon atoms of a straight, branched, or cyclic  
configuration. Examples of alkylcarbonyl groups are  
2-methylbutanoyl, octadecanoyl, 11-cyclohexyl-  
25   undecanoyl and the like. Thus, the 11-cyclohexyl-  
undecanoyl group is c-Hex-(CH<sub>2</sub>)<sub>10</sub>-C(O)-.

          The term "lower alkylcarbonyl" means  
alkylcarbonyl groups of from 1 to 8 carbon atoms of a  
straight, branched, or cyclic configuration.  
30   Examples of lower alkylcarbonyl groups are formyl,  
2-methylbutanoyl, cyclohexylacetyl, etc. By way of  
illustration, the 2-methylbutanoyl groups signifies  
-C(O)CH(CH<sub>3</sub>)CH<sub>2</sub>CH<sub>3</sub>.

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Substituted-phenyl, -benzyl, -2-phenethyl, or -pyridinyl means that the aromatic ring carries 1 or 2 substituents selected from lower alkyl,  $R^{10}$ ,  $NO_2$ ,  $SCF_3$ , halogen,  $-C(O)R^7$ ,  $-C(O)R^{10}$ , CN,  $CF_3$ , and Tz.

Halogen includes F, Cl, Br, and I.

It is intended that the definitions of any substituent (e.g.,  $R^1$ ,  $R^2$ ,  $R^{10}$ , etc.) in a particular molecule be independent of its definitions elsewhere in the molecule. Thus,  $-NR^{12}R^{12}$  represents  $-NHH$ ,  $-NHCH_3$ ,  $-NHC_6H_5$ , etc.

The saturated rings formed when two  $R^1$  groups join through two adjacent carbon atoms include c-pentane, c-hexane, c-heptane, c-octane, c-nonane, and c-decane.

The saturated rings formed when two  $R^2$  groups join through C include c-propane, c-pentane, c-hexane, c-octane, tetrahydrofuran, tetrahydrothiophene, pyrrolidine, pyran, thiopyran, piperidine, dioxane, morpholine, thiomorpholine, piperazine, and their N-lower alkyl analogs.

The heterocycles formed when two  $R^{12}$  or  $R^{20}$  groups join through N include pyrrolidine, piperidine, morpholine, thiamorpholine, piperazine, and N-methylpiperazine.

When  $Q^1$  and  $R^{22}$  and the carbons through which they are attached form a ring, the rings thus formed include lactones, lactams, and thiolactones.

The prodrug esters of  $Q^1$  (i.e., when  $Q^1 = CO_2R^6$ ) are intended to include the esters such as are described by Saari *et al.*, J. Med. Chem., 21, No. 8, 746-753 (1978), Sakamoto *et al.*, Chem. Pharm. Bull., 32, No. 6, 2241-2248 (1984), and Bundgaard *et al.*, J. Med. Chem., 30, No. 3, 451-454 (1987).

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Within the definition of  $R^8$ , some representative monocyclic or bicyclic heterocyclic radicals are:

- 5        2,5-dioxo-1-pyrrolidinyl,  
         (3-Pyridinylcarbonyl)amino,  
         1,3-dihydro-1,3-dioxo-2H-isoindol-2-yl,  
         1,3-dihydro-2H-isoindol-2-yl,  
         2,4-imidazolin-1-yl,  
10       2,6-piperidinedion-1-yl,  
         2-imidazolyl,  
         2-oxo-1,3-dioxolen-4-yl,  
         piperidin-1-yl,  
         morpholin-1-yl, and  
15       piperazin-1-yl.

- "Standard amino acid", the radical of which may be  $CR^3R^{22}$ , means the following amino acids:  
alanine, asparagine, aspartic acid, arginine,  
20       cysteine, glutamic acid, glutamine, glycine,  
         histidine, isoleucine, leucine, lysine, methionine,  
         phenylalanine, proline, serine, threonine,  
         tryptophan, tyrosine, and valine. See F.H.C. Crick,  
         Symposium of the Society of Experimental Biology, 12,  
25       140 (1958).

Optical Isomers - Diastereomers - Geometric Isomers

- Some of the compounds described herein contain one or more asymmetric centers and may thus  
30       give rise to diastereomers and optical isomers. The  
         present invention is meant to comprehend such  
         possible diastereomers as well as their racemic and

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resolved, enantiomerically pure forms and pharmaceutically acceptable salts thereof. Optically active (R) and (S) isomers may be resolved using conventional techniques.

5           Some of the compounds described herein contain olefinic double bonds, and unless specified otherwise, are meant to include both E and Z geometric isomers.

10    Salts

          The pharmaceutical compositions of the present invention comprise a compound of Formula I as an active ingredient or a pharmaceutically acceptable salt, thereof, and may also contain a pharmaceutically acceptable carrier and optionally other therapeutic ingredients. The term "pharmaceutically acceptable salts" refers to salts prepared from pharmaceutically acceptable non-toxic bases including inorganic bases and organic bases. Salts derived from inorganic bases include aluminum, ammonium, calcium, copper, ferric, ferrous, lithium, magnesium, manganic salts, manganous, potassium, sodium, zinc, and the like. Particularly preferred are the ammonium, calcium, magnesium, potassium, and sodium salts. Salts derived from pharmaceutically acceptable organic non-toxic bases include salts of primary, secondary, and tertiary amines, substituted amines including naturally occurring substituted amines, cyclic amines, and basic ion exchange resins, such as arginine, betaine, caffeine, choline, N,N'-dibenzylethylenediamine, diethylamine, 2-diethylaminoethanol, 2-dimethylaminoethanol,

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ethanolamine, ethylenediamine, N-ethylmorpholine, N-ethylpiperidine, glucamine, glucosamine, histidine, hydrabamine, isopropylamine, lysine, methylglucamine, morpholine, piperazine, piperidine, polyamine resins, procaine, purines, theobromine, triethylamine, trimethylamine, tripropylamine, tromethamine, and the like.

When the compound of the present invention is basic, salts may be prepared from pharmaceutically acceptable non-toxic acids, including inorganic and organic acids. Such acids include acetic, benzenesulfonic, benzoic, camphorsulfonic, citric, ethanesulfonic, fumaric, gluconic, glutamic, hydrobromic, hydrochloric, isethionic, lactic, maleic, malic, mandelic, methanesulfonic, mucic, nitric, pamoic, pantothenic, phosphoric, succinic, sulfuric, tartaric, p-toluenesulfonic acid, and the like. Particularly preferred are citric, hydrobromic, hydrochloric, maleic, phosphoric, sulfuric, and tartaric acids.

It will be understood that in the discussion of methods of treatment which follows, references to the compounds of Formula I are meant to also include the pharmaceutically acceptable salts.

25

### Utilities

The ability of the compounds of Formula I to antagonize the actions of the leukotrienes makes them useful for preventing or reversing the symptoms induced by the leukotrienes in a human subject. This antagonism of the actions of leukotrienes indicates that the compounds and pharmaceutical compositions

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thereof are useful to treat, prevent, or ameliorate in mammals and especially in humans: 1) pulmonary disorders including diseases such as asthma, chronic bronchitis, and related obstructive airway diseases, 2) allergies and allergic reactions such as allergic rhinitis, contact dermatitis, allergic conjunctivitis, and the like, 3) inflammation such as arthritis or inflammatory bowel disease, 4) pain, 5) skin disorders such as psoriasis, atopic eczema, and the like, 6) cardiovascular disorders such as angina, myocardial ischemia, hypertension, platelet aggregation and the like, 7) renal insufficiency arising from ischaemia induced by immunological or chemical (cyclosporin) etiology, 8) migraine or cluster headache, 9) ocular conditions such as uveitis, 10) hepatitis resulting from chemical, immunological or infectious stimuli, 11) trauma or shock states such as burn injuries, endotoxemia and the like, 12) allograft rejection, 13) prevention of side effects associated with therapeutic administration of cytokines such as Interleukin II and tumor necrosis factor, 14) chronic lung diseases such as cystic fibrosis, bronchitis and other small- and large-airway diseases, and 15) cholecystitis.

Thus, the compounds of the present invention may also be used to treat or prevent mammalian (especially, human) disease states such as erosive gastritis; erosive esophagitis; diarrhea; cerebral spasm; premature labor; spontaneous abortion; dysmenorrhea; ischemia; noxious agent-induced damage or necrosis of hepatic, pancreatic, renal, or myocardial tissue; liver parenchymal damage caused by

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hepatotoxic agents such as  $\text{CCl}_4$  and D- galactosamine; ischemic renal failure; disease-induced hepatic damage; bile salt induced pancreatic or gastric damage; trauma- or stress-induced cell damage; and  
5 glycerol-induced renal failure. The compounds also exhibit cytoprotective action.

The cytoprotective activity of a compound may be observed in both animals and man by noting the increased resistance of the gastrointestinal mucosa  
10 to the noxious effects of strong irritants, for example, the ulcerogenic effects of aspirin or indomethacin. In addition to lessening the effect of non-steroidal anti-inflammatory drugs on the gastrointestinal tract, animal studies show that  
15 cytoprotective compounds will prevent gastric lesions induced by oral administration of strong acids, strong bases, ethanol, hypertonic saline solutions, and the like.

Two assays can be used to measure  
20 cytoprotective ability. These assays are; (A) an ethanol-induced lesion assay and (B) an indomethacin-induced ulcer assay and are described in EP 140,684.

#### 25 Dose Ranges

The magnitude of prophylactic or therapeutic dose of a compound of Formula I will, of course, vary with the nature of the severity of the condition to  
- be treated and with the particular compound of  
30 Formula I and its route of administration. It will also vary according to the age, weight and response of the individual patient. In general, the daily

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dose range for anti-asthmatic, anti-allergic or anti-inflammatory use and generally, uses other than cytoprotection, lie within the range of from about 0.001 mg to about 100 mg per kg body weight of a mammal, preferably 0.01 mg to about 10 mg per kg, and most preferably 0.1 to 1 mg per kg, in single or divided doses. On the other hand, it may be necessary to use dosages outside these limits in some cases.

For use where a composition for intravenous administration is employed, a suitable dosage range for anti-asthmatic, anti-inflammatory, or anti-allergic use is from about 0.001 mg to about 25 mg (preferably from 0.01 mg to about 1 mg) of a compound of Formula I per kg of body weight per day and for cytoprotective use from about 0.1 mg to about 100 mg (preferably from about 1 mg to about 100 mg and more preferably from about 1 mg to about 10 mg) of a compound of Formula I per kg of body weight per day.

In the case where an oral composition is employed, a suitable dosage range for anti-asthmatic, anti-inflammatory or anti-allergic use is, e.g. from about 0.01 mg to about 100 mg of a compound of Formula I per kg of body weight per day, preferably from about 0.1 mg to about 10 mg per kg and for cytoprotective use from 0.1 mg to about 100 mg (preferably from about 1 mg to about 100 mg and more preferably from about 10 mg to about 100 mg) of a compound of Formula I per kg of body weight per day.

For the treatment of diseases of the eye, ophthalmic preparations for ocular administration comprising 0.001-1% by weight solutions or

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suspensions of the compounds of Formula I in an acceptable ophthalmic formulation may be used.

The exact amount of a compound of the Formula I to be used as a cytoprotective agent will depend on, inter alia, whether it is being administered to heal damaged cells or to avoid future damage, on the nature of the damaged cells (e.g., gastrointestinal ulcerations vs. nephrotic necrosis), and on the nature of the causative agent. An example of the use of a compound of the Formula I in avoiding future damage would be co-administration of a compound of the Formula I with an NSAID that might otherwise cause such damage (for example, indomethacin). For such use, the compound of Formula I is administered from 30 minutes prior up to 30 minutes after administration of the NSAID. Preferably it is administered prior to or simultaneously with the NSAID, (for example, in a combination dosage form).

#### Pharmaceutical Compositions

Any suitable route of administration may be employed for providing a mammal, especially a human with an effective dosage of a compound of the present invention. For example, oral, rectal, topical, parenteral, ocular, pulmonary, nasal, and the like may be employed. Dosage forms include tablets, troches, dispersions, suspensions, solutions, capsules, creams, ointments, aerosols, and the like.

The pharmaceutical compositions of the present invention comprise a compound of Formula I as an active ingredient or a pharmaceutically acceptable

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salt thereof, and may also contain a pharmaceutically acceptable carrier and optionally other therapeutic ingredients. The term "pharmaceutically acceptable salts" refers to salts prepared from pharmaceutically acceptable non-toxic bases or acids including inorganic bases or acids and organic bases or acids.

The compositions include compositions suitable for oral, rectal, topical, parenteral (including subcutaneous, intramuscular, and intravenous), ocular (ophthalmic), pulmonary (nasal or buccal inhalation), or nasal administration, although the most suitable route in any given case will depend on the nature and severity of the conditions being treated and on the nature of the active ingredient. They may be conveniently presented in unit dosage form and prepared by any of the methods well-known in the art of pharmacy.

For administration by inhalation, the compounds of the present invention are conveniently delivered in the form of an aerosol spray presentation from pressurized packs or nebulisers. The compounds may also be delivered as powders which may be formulated and the powder composition may be inhaled with the aid of an insufflation powder inhaler device. The preferred delivery system for inhalation is a metered dose inhalation (MDI) aerosol, which may be formulated as a suspension or solution of a compound of Formula I in suitable propellants, such as fluorocarbons or hydrocarbons.

Suitable topical formulations of a compound of formula I include transdermal devices, aerosols, creams, ointments, lotions, dusting powders, and the like.

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In practical use, the compounds of Formula I can be combined as the active ingredient in intimate admixture with a pharmaceutical carrier according to conventional pharmaceutical compounding techniques.

5 ~~The carrier may take a wide variety of forms~~  
depending on the form of preparation desired for  
administration, e.g., oral or parenteral (including  
intravenous). In preparing the compositions for oral  
dosage form, any of the usual pharmaceutical media  
10 may be employed, such as, for example, water,  
glycols, oils, alcohols, flavoring agents,  
preservatives, coloring agents and the like in the  
case of oral liquid preparations, such as, for  
example, suspensions, elixirs and solutions; or  
15 carriers such as starches, sugars, microcrystalline  
cellulose, diluents, granulating agents, lubricants,  
binders, disintegrating agents and the like in the  
case of oral solid preparations such as, for example,  
powders, capsules and tablets, with the solid oral  
20 preparations being preferred over the liquid  
preparations. Because of their ease of  
administration, tablets and capsules represent the  
most advantageous oral dosage unit form in which case  
solid pharmaceutical carriers are obviously  
25 employed. If desired, tablets may be coated by  
standard aqueous or nonaqueous techniques.

In addition to the common dosage forms set  
out above, the compounds of Formula I may also be  
30 ~~administered by controlled release means and/or~~  
delivery devices such as those described in U.S.  
Patent Nos. 3,845,770; 3,916,899; 3,536,809;  
3,598,123; 3,630,200 and 4,008,719, the disclosures  
of which are hereby incorporated herein by reference.

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Pharmaceutical compositions of the present invention suitable for oral administration may be presented as discrete units such as capsules, cachets or tablets each containing a predetermined amount of the active ingredient, as a powder or granules or as a solution or a suspension in an aqueous liquid, a non-aqueous liquid, an oil-in-water emulsion or a water-in-oil liquid emulsion. Such compositions may be prepared by any of the methods of pharmacy but all methods include the step of bringing into association the active ingredient with the carrier which constitutes one or more necessary ingredients. In general, the compositions are prepared by uniformly and intimately admixing the active ingredient with liquid carriers or finely divided solid carriers or both, and then, if necessary, shaping the product into the desired presentation. For example, a tablet may be prepared by compression or molding, optionally with one or more accessory ingredients. Compressed tablets may be prepared by compressing in a suitable machine, the active ingredient in a free-flowing form such as powder or granules, optionally mixed with a binder, lubricant, inert diluent, surface active or dispersing agent. Molded tablets may be made by molding in a suitable machine, a mixture of the powdered compound moistened with an inert liquid diluent. Desirably, each tablet contains from about 2.5 mg to about 500 mg of the active ingredient and each cachet or capsule contains from about 2.5 to about 500 mg of the active ingredient.

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The following are examples of representative pharmaceutical dosage forms for the compounds of Formula I:

5	<u>Injectable Suspension (I.M.)</u>	<u>mg/mL</u>
	Compound of Formula I	10
	Methylcellulose	5.0
	Tween 80	0.5
	Benzyl alcohol	9.0
10	Benzalkonium chloride	1.0
	Water for injection to a total volume of 1 mL	
	<u>Tablet</u>	<u>mg/tablet</u>
	Compound of Formula I	25
15	Microcrystalline Cellulose	415
	Povidone	14.0
	Pregelatinized Starch	43.5
	Magnesium Stearate	<u>2.5</u>
		500
20	<u>Capsule</u>	<u>mg/capsule</u>
	Compound of Formula I	25
	Lactose Powder	573.5
	Magnesium Stearate	<u>1.5</u>
25		600
	<u>Aerosol</u>	<u>Per canister</u>
	Compound of Formula I	24 mg
	<del>Lecithin, NF Liquid Concentrate</del>	1.2 mg
30	Trichlorofluoromethane, NF	4.025 g
	Dichlorodifluoromethane, NF	12.15 g

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Combinations With Other Drugs

In addition to the compounds of Formula I, the pharmaceutical compositions of the present invention can also contain other active ingredients, such as cyclooxygenase inhibitors, non-steroidal anti-inflammatory drugs (NSAIDs), peripheral analgesic agents such as zomepirac, diflunisal and the like. The weight ratio of the compound of the Formula I to the second active ingredient may be varied and will depend upon the effective dose of each ingredient. Generally, an effective dose of each will be used. Thus, for example, when a compound of the Formula I is combined with an NSAID the weight ratio of the compound of the Formula I to the NSAID will generally range from about 1000:1 to about 1:1000, preferably about 200:1 to about 1:200. Combinations of a compound of the Formula I and other active ingredients will generally also be within the aforementioned range, but in each case, an effective dose of each active ingredient should be used.

NSAIDs can be characterized into five groups:

- (1) propionic acid derivatives;
  - (2) acetic acid derivatives;
  - (3) fenamic acid derivatives;
  - (4) oxicams; and
  - (5) biphenylcarboxylic acid derivatives,
- or a pharmaceutically acceptable salt thereof.

The propionic acid derivatives which may be used comprise: alminoprofen, benoxaprofen, bucloxic acid, carprofen, fenbufen, fenoprofen, fluprofen, flurbiprofen, ibuprofen, indoprofen, ketoprofen, miroprofen, naproxen, oxaprozin, piroprofen,

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prano-profen, suprofen, tiaprofenic acid, and  
tioxaprofen. Structurally related propionic acid  
derivatives having similar analgesic and anti-  
inflammatory properties are also intended to be  
5 included in this group.

Thus, "propionic acid derivatives" as  
defined herein are non-narcotic analgesics/  
non-steroidal anti-inflammatory drugs having a free  
-CH(CH<sub>3</sub>)COOH or -CH<sub>2</sub>CH<sub>2</sub>COOH group (which optionally  
10 can be in the form of a pharmaceutically acceptable  
salt group, e.g., -CH(CH<sub>3</sub>)COO<sup>-</sup>Na<sup>+</sup> or -CH<sub>2</sub>CH<sub>2</sub>COO<sup>-</sup>Na<sup>+</sup>),  
typically attached directly or via a carbonyl  
function to a ring system, preferably to an aromatic  
ring system.

15 The acetic acid derivatives which may be  
used comprise: indomethacin, which is a preferred  
NSAID, acetaminophen, alclometacin, clidametacin, diclofenac,  
fenclofenac, fenclozic acid, fentiazac, furofenac,  
ibufenac, isoxepac, oxpinac, sulindac, tiopinac,  
20 tolmetin, zidometacin, and zomepirac. Structurally  
related acetic acid derivatives having similar  
analgesic and anti-inflammatory properties are also  
intended to be encompassed by this group.

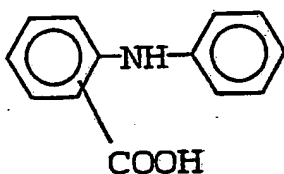
Thus, "acetic acid derivatives" as defined  
25 herein are non-narcotic analgesics/non-steroidal  
anti-inflammatory drugs having a free -CH<sub>2</sub>COOH group  
(which optionally can be in the form of a  
pharmaceutically acceptable salt group, e.g.  
-CH<sub>2</sub>COO<sup>-</sup>Na<sup>+</sup>), typically attached directly to a ring  
30 system, preferably to an aromatic or heteroaromatic  
ring system.

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The fenamic acid derivatives which may be used comprise: flufenamic acid, meclofenamic acid, mefenamic acid, niflumic acid and tolfenamic acid. Structurally related fenamic acid derivatives having similar analgesic and anti-inflammatory properties are also intended to be encompassed by this group.

Thus, "fenamic acid derivatives" as defined herein are non-narcotic analgesics/non-steroidal anti-inflammatory drugs which contain the basic structure:

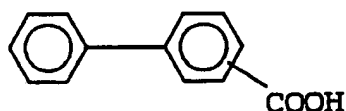


which can bear a variety of substituents and in which the free -COOH group can be in the form of a pharmaceutically acceptable salt group, e.g., -COO<sup>-</sup>Na<sup>+</sup>.

The biphenylcarboxylic acid derivatives which can be used comprise: diflunisal and flufenisal. Structurally related biphenylcarboxylic acid derivatives having similar analgesic and anti-inflammatory properties are also intended to be encompassed by this group.

Thus, "biphenylcarboxylic acid derivatives" as defined herein are non-narcotic analgesics/non-steroidal anti-inflammatory drugs which contain the basic structure:

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5

which can bear a variety of substituents and in which the free -COOH group can be in the form of a pharmaceutically acceptable salt group, e.g., -COO<sup>-</sup>Na<sup>+</sup>.

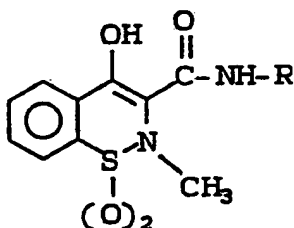
10

The oxicams which can be used in the present invention comprise: isoxicam, piroxicam, sudoxicam and tenoxicam. Structurally related oxicams having similar analgesic and anti-inflammatory properties are also intended to be encompassed by this group.

15

Thus, "oxicams" as defined herein are non-narcotic analgesics/non-steroidal anti-inflammatory drugs which have the general formula:

20



25

wherein R is an aryl or heteroaryl ring system.

The following NSAIDs may also be used:

amfenac sodium, aminoprofen, anitrazafen,

antrafenine, auranofin, bendazac lysinate,

30

benzydanine, beprozin, broperamole, bufezolac,

cinmetacin, ciproquazone, cloximate, dazidamine,

deboxamet, delmetacin, detomidine, dexindoprofen,

diacerein, di-fisalamine, difenpyramide, emorfazone,

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5 enfenamic acid, enolicam, epirizole, etersalate,  
etodolac, etofenamate, fanetizole mesylate,  
fenclorac, fendosal, fenflumizole, feprazone,  
floctafenine, flunixin, flunoxaprofen, fluproquazone,  
5 fopirtoline, fosfosal, furclopofen, glucametacin,  
guaimesal, ibuproxam, isofezolac, isonixim,  
isoprofen, isoxicam, lefetamine HCl, leflunomide,  
lofemizole, lonazolac calcium, lotifazole,  
loxoprofen, lysin clonixinate, meclofenamate sodium,  
10 meseclazone, nabumetone, nictindole, nimesulide,  
orpanoxin, oxametacin, oxapadol, perisoxal citrate,  
pimeprofen, pimetacin, piroxan, pirazolac,  
pirfenidone, proglumetacin maleate, proquazone,  
pyridoxiprofen, sudoxicam, talmetacin, talniflumate,  
15 tenoxicam, thiazolinobutazone, thielavin B, tiaramide  
HCl, tiflamizole, timegadine, tolpadol, tryptamid,  
and ufenamate.

The following NSAIDs, designated by company  
code number (see e.g., Pharmaprojects), may also be  
20 used: 480156S, AA861, AD1590, AEP802, AEP860, AI77B,  
AP504, AU8001, BPPC, BW540C, CHINOIN 127, CN100,  
EB382, EL508, F1044, GV3658, ITF182, KCNTEI6090,  
KME4, LA2851, MR714, MR897, MY309, ONO3144, PR823,  
PV102, PV108, R830, RS2131, SCR152, SH440, SIR133,  
25 SPAS510, SQ27239, ST281, SY6001, TA60, TAI-901  
(4-benzoyl-1-indancarboxylic acid), TVX2706, U60257,  
UR2301, and WY41770.

Finally, NSAIDs which may also be used  
include the salicylates, specifically acetyl  
30 salicylic acid and the phenylbutazones, and  
pharmaceutically acceptable salts thereof.

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In addition to indomethacin, other preferred NSAIDs are acetyl salicylic acid, diclofenac, fenbufen, fenoprofen, flurbiprofen, ibuprofen, ketoprofen, naproxen, phenylbutazone, piroxicam, sulindac, and tolmetin.

Pharmaceutical compositions comprising the Formula I compounds may also contain inhibitors of the biosynthesis of the leukotrienes such as are disclosed in EP 138,481 (April 24, 1985), EP 115,394 (August 8, 1984), EP 136,893 (April 10, 1985), and EP 140,709 (May 8, 1985), which are hereby incorporated herein by reference.

The compounds of the Formula I may also be used in combination with leukotriene antagonists such as those disclosed in EP 106,565 (April 25, 1984) and EP 104,885 (April 4, 1984) which are hereby incorporated herein by reference and others known in the art such as those disclosed in EP Application Nos. 56,172 (July 21, 1982) and 61,800 (June 10, 1982); and in U.K. Patent Specification No. 2,058,785 (April 15, 1981), which are hereby incorporated herein by reference.

Pharmaceutical compositions comprising the Formula I compounds may also contain as the second active ingredient, prostaglandin antagonists such as those disclosed in EP 11,067 (May 28, 1980) or thromboxane antagonists such as those disclosed in U.S. Pat. 4,237,160. They may also contain histidine decarboxylase inhibitors such as  $\alpha$ -fluoromethyl-histidine, described in U.S. Pat. 4,325,961. The compounds of the Formula I may also be advantageously combined with an H<sub>1</sub>- or H<sub>2</sub>-receptor antagonist, such

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as for instance acetamazole, aminothiadiazoles disclosed in EP 40,696 (December 2, 1981), benadryl, cimetidine, famotidine, framamine, histadyl, phenergan, ranitidine, terfenadine and like compounds, such as those disclosed in U.S. Patent Nos. 4,283,408; 4,362,736; and 4,394,508. The pharmaceutical compositions may also contain a  $K^+/H^+$  ATPase inhibitor such as omeprazole, disclosed in U.S. Pat. 4,255,431, and the like. Compounds of Formula I may also be usefully combined with most cell stabilizing agents, such as 1,3-bis(2-carboxy-chromon-5-yloxy)-2-hydroxypropane and related compounds described in British Patent Specifications 1,144,905 and 1,144,906. Another useful pharmaceutical composition comprises the Formula I compounds in combination with serotonin antagonists such as methysergide, the serotonin antagonists described in Nature, 316, 126-131 (1985), and the like. Each of the references referred to in this paragraph is hereby incorporated herein by reference.

Other advantageous pharmaceutical compositions comprise the Formula I compounds in combination with anti-cholinergics such as ipratropium bromide, bronchodilators such as the beta agonist salbutamol, metaproterenol, terbutaline, fenoterol and the like, and the anti-asthmatic drugs theophylline, choline theophyllinate and enprofylline, the calcium antagonists nifedipine, diltiazem, nitrendipine, verapamil, nimodipine, felodipine, etc. and the corticosteroids, hydrocortisone, methylprednisolone, betamethasone, dexamethasone, beclomethasone, and the like.

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Methods Of Synthesis

Compounds of the present invention can be prepared according to the following methods.

5      Scheme 1

2-Methylpyridine of general structure III is prepared by  $\alpha$ -methylation of substituted pyridine II using the method of Kray and Reinecke (J. Am. Chem. Soc., 1964, 86, 5355). Oxidation with  
10 m-chloroperbenzoic acid gives substituted 2-methylpyridine N-oxide IV. Treatment of the latter with phosphoryl chloride gives the 2-chloromethyl substituted pyridine VII. Alternatively, refluxing IV in acetic anhydride gives the 2-(acetoxymethyl)  
15 pyridine VII which is hydrolysed with aqueous base to the corresponding alcohol VII. When this hydroxymethyl derivative is treated with methanesulfonyl chloride in the presence of triethylamine, the 2-(methanesulfonyloxymethyl)pyridine is obtained,  
20 whereas treatment with brominetriphenylphosphine gives the 2-(bromomethyl)pyridine VII. When either the bromomethyl, chloromethyl, or the mesyloxymethyl pyridine derivative VII is refluxed in acetonitrile in the presence of triphenylphosphine, the  
25 corresponding phosphonium salt VIII is formed.

An alternative route to the 2-(hydroxymethyl) pyridine VII from variously substituted pyridine II involves (1) oxidation to the pyridine N-oxide, (2) a modified Reissert reaction according to the protocol  
30 of W.F. Fife (J. Org. Chem., 1983, 48, 1375) to give the 2-cyano pyridine VI, (3) conversion to the methyl ester with anhydrous methanol and acid, and (4) finally reduction of the 2-carbomethoxyppyridine with diisobutylaluminium hydride.

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Scheme 2

2-Pyridinecarboxaldehydes such as IX  
(obtained by an oxidative procedure from VI1 or  
partial reduction of VI) can be deprotonated at the  
5 6-benzylic position by successive treatment with  
lithium N,N,N-trimethylethylenediamine (LiTMEDA) and  
lithium diisopropylamide. Addition of an alkylating  
agent R'X gives the pyridine derivative X with an  
alkyl substituent at the 6-position. Reduction with  
10 sodium borohydride gives the carbinol XI which is  
then transformed into the phosphonium salt XII as  
described in Scheme 1.

Ramification of the 5-alkyl substituent of a  
2-pyridinecarbonitrile XIII is done by treatment with  
15 a base such as potassium hexamethyldisilazide  
followed by an alkylating agent R''X to give XIV. The  
latter is converted to the phosphonium salt XV as  
described in Scheme 1.

20 Scheme 3

When treated with various organometallic  
reagents ( $R^2M$ ), with or without catalyst,  
2-halopyridines of general structure XVI give the  
corresponding 2-substituted pyridines XVII (such as  
25 in Example 2). The latter are also prepared by  
acylation of the pyridine N-oxide V followed by  
treatment with an organometallic reagent  $R^2M$ .  
Alternatively, condensation of enamines XVIII with an  
-  $\alpha,\beta$ -unsaturated carbonyl gives a dicarbonyl  
30 intermediate XIX which is cyclised to the pyridine XX  
with ammonia or an ammonium salt.

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Scheme 4

Dialdehyde XXI is reduced with sodium borohydride. The resulting alcohol is protected as its tetrahydropyranyl ether XXII which is then  
5 treated with vinyl magnesium bromide or allyl magnesium bromide to give the alcohol XXIII. Coupling of XXIII with bromide XXIV in the presence of palladium acetate using the procedure of R.C. Larock et al. (Tetrahedron Letters, 30, 6629 (1989))  
10 gives the keto ester XXV. The ketone is then reduced using the chiral oxazaborolidine complex with borane XXVI (J. Org. Chem., 56, 751, 1991) followed by reaction of the ester with an alkyl Grignard and cerium chloride to give the diol XXVII. (To obtain  
15 compound XXVII with one  $R^2 = H$  cerium chloride is omitted, one equivalent of Grignard is used and the initially formed ketone is reduced to the corresponding benzyl alcohol.)

The chiral alcohol of the diol XXVII is  
20 first protected as its t-butyldimethylsilyl ether. The other benzylic alcohol is protected as a tetrahydropyranyl ether which is then treated with tetrabutylammonium fluoride to give the alcohol XXVIII. Mesylation of XXVIII followed by  
25 displacement of the resulting mesylate with the appropriate substituted thiol XXIX gives the thioether XXX protected as a tetrahydropyranyl ether. The hydroxythioether XXX is obtained directly from diol XXVII by mesylation followed by thiol  
30 displacement.

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Deprotection of the benzylic alcohol(s) of XXX and oxidation gives the benzaldehyde XXXI. Coupling of XXXI with VIII gives the olefin linked pyridine benzyl alcohol XXXII (I). In the case where  
5 ~~Q<sup>1</sup> is an ester, hydrolysis with a base such as NaOH or LiOH (followed by acidification) affords the corresponding acid XXXII (I).~~

It will be obvious to one skilled in the art that compound XXXII (I) having the opposite  
10 stereochemistry at the sulfur-bearing benzylic carbon can be obtained by using the opposite enantiomer of the chiral reduction catalyst XXVI to reduce XXV to XXVII or by inversion of the stereocenter in XXVIII by a Mitsunobu reaction (Synthesis, 1-28, 1981).

15

Scheme 5

Reduction of ketoaldehyde XXXIII, followed by protection of the corresponding ketoalcohol as its tetrahydropyranyl ether gives XXXIV. The enolate of  
20 ketone XXXIV, obtained by treatment of XXXIV with a base such as KH or NaH, is reacted with dimethylcarbonate to yield the keto ester XXXV. Alkylation of the keto ester XXXV with iodide XXXVI followed by decarboxylation of the resulting adduct using  
25 conditions such as heating with HCl in acetic acid affords the ketone XXXVII. In the case where the THP ether is cleaved, the alcohol is reprotected as the THP ether. Following the procedure described in Scheme 4, ketone XXXVII is transformed to XXXVIII, a  
30 structure representative of I.

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Scheme 6

Iodoacid XXXIX is treated with 2 equivalents of a base such as n-butyllithium in a suitable solvent such as THF at -100°C, then at -78°C to  
5 afford XL, which is reacted with aldehyde XXI to yield the hydroxyacid XLI. The acid XLI is esterified using conditions such as CH<sub>2</sub>N<sub>2</sub> or CH<sub>3</sub>I/Cs<sub>2</sub>CO<sub>3</sub> and an organometallic reagent is then added to give the diol XLII. Following the same procedure as  
10 in Scheme 4, the diol XLII is transformed to XLIII, which is a structure representative of I.

Scheme 7

Treatment of olefin-linked pyridine XXXII with trimethylsulfonium iodide gives the cyclopropyl-linked compound XLIV, which is a representative of I. Reduction of the olefin of XXXII with borane gives the saturated compound XLV, which is another  
20 representative of I.

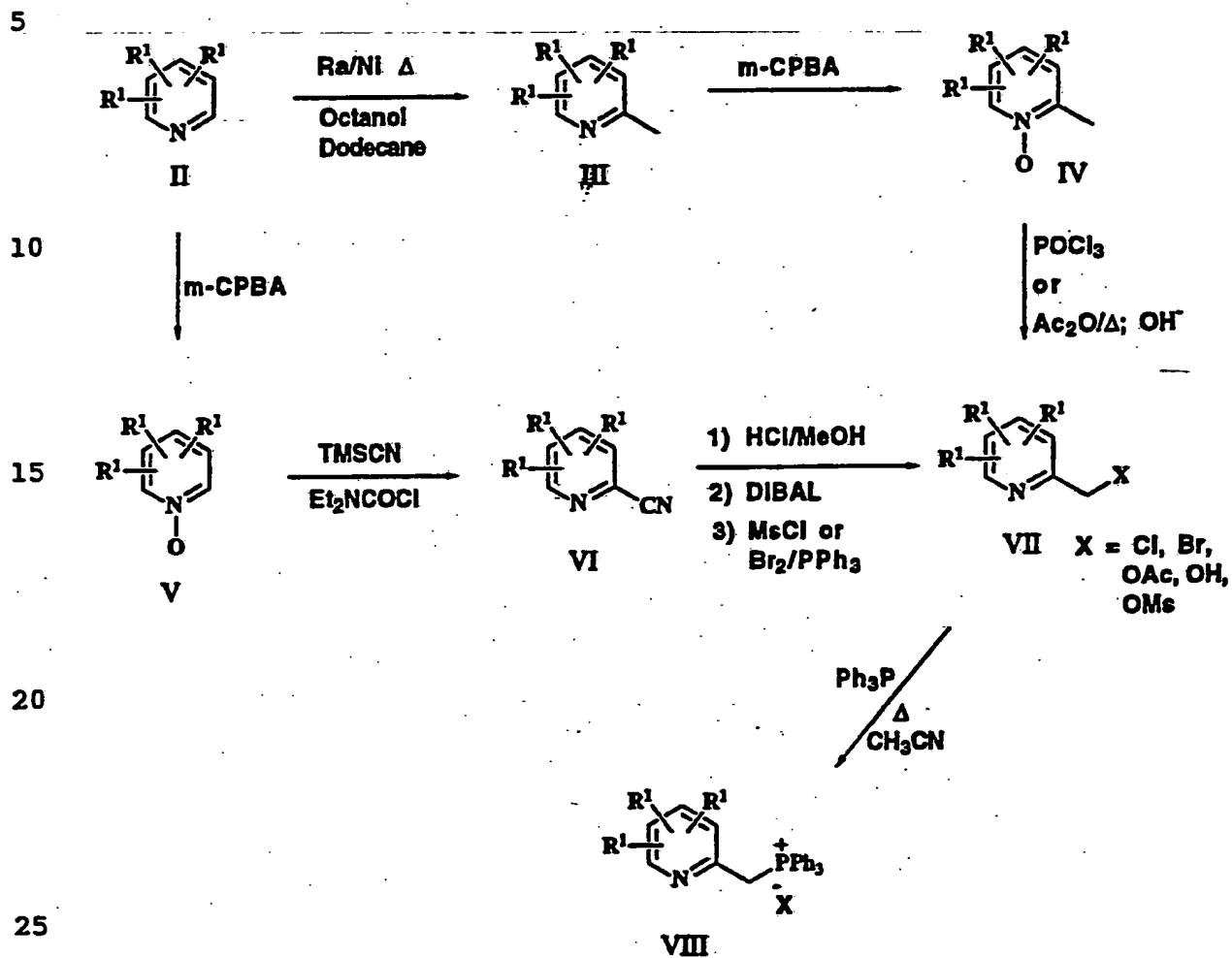
Scheme 8

Starting from benzaldehyde derivative XLVI and following the same sequence as described in Scheme 4, XLVII is prepared, which is coupled with  
25 the halide VII to give the ether and the thioether linked pyridine XLVIII, which is a representative of I.

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SCHEME 1

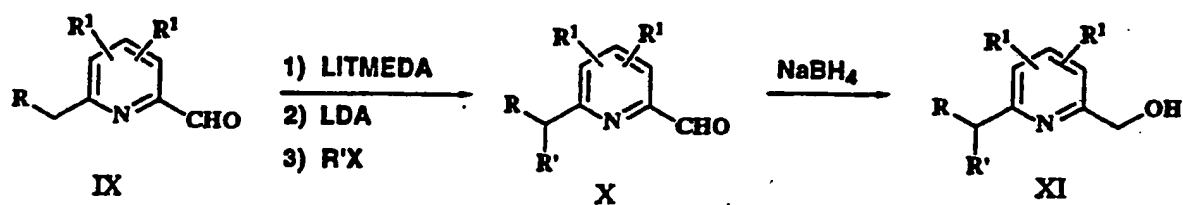
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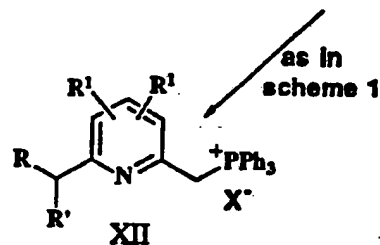
SCHEME 2

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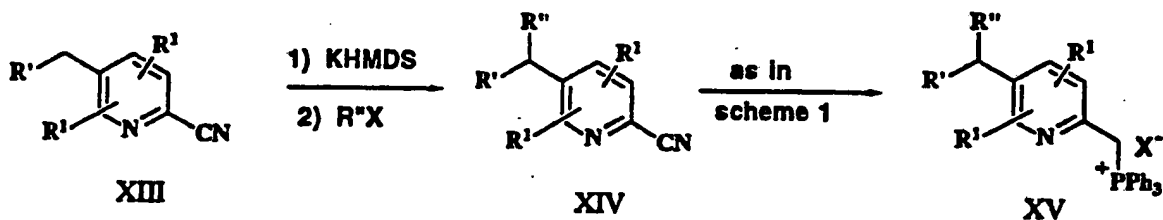
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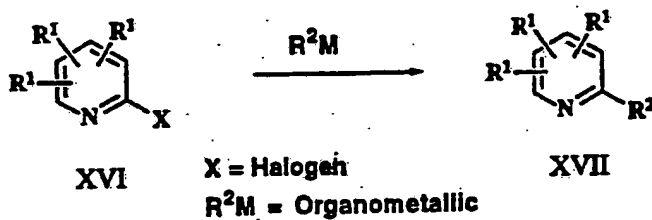
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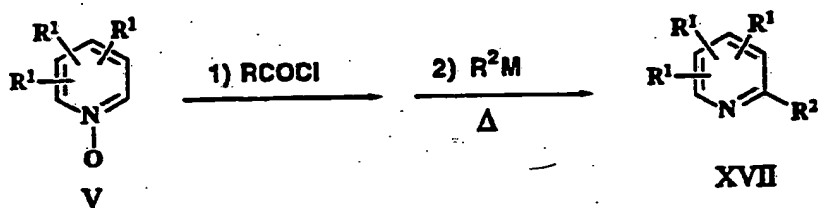
## SCHEME 3

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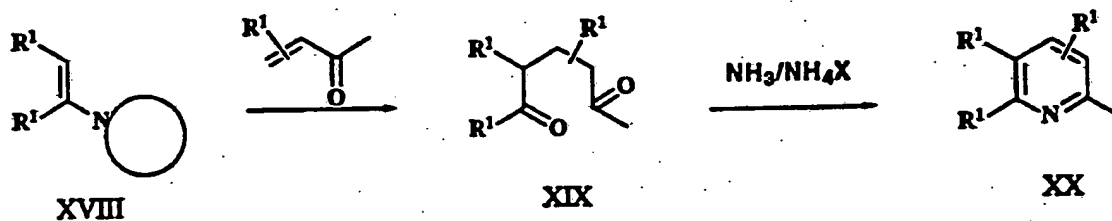


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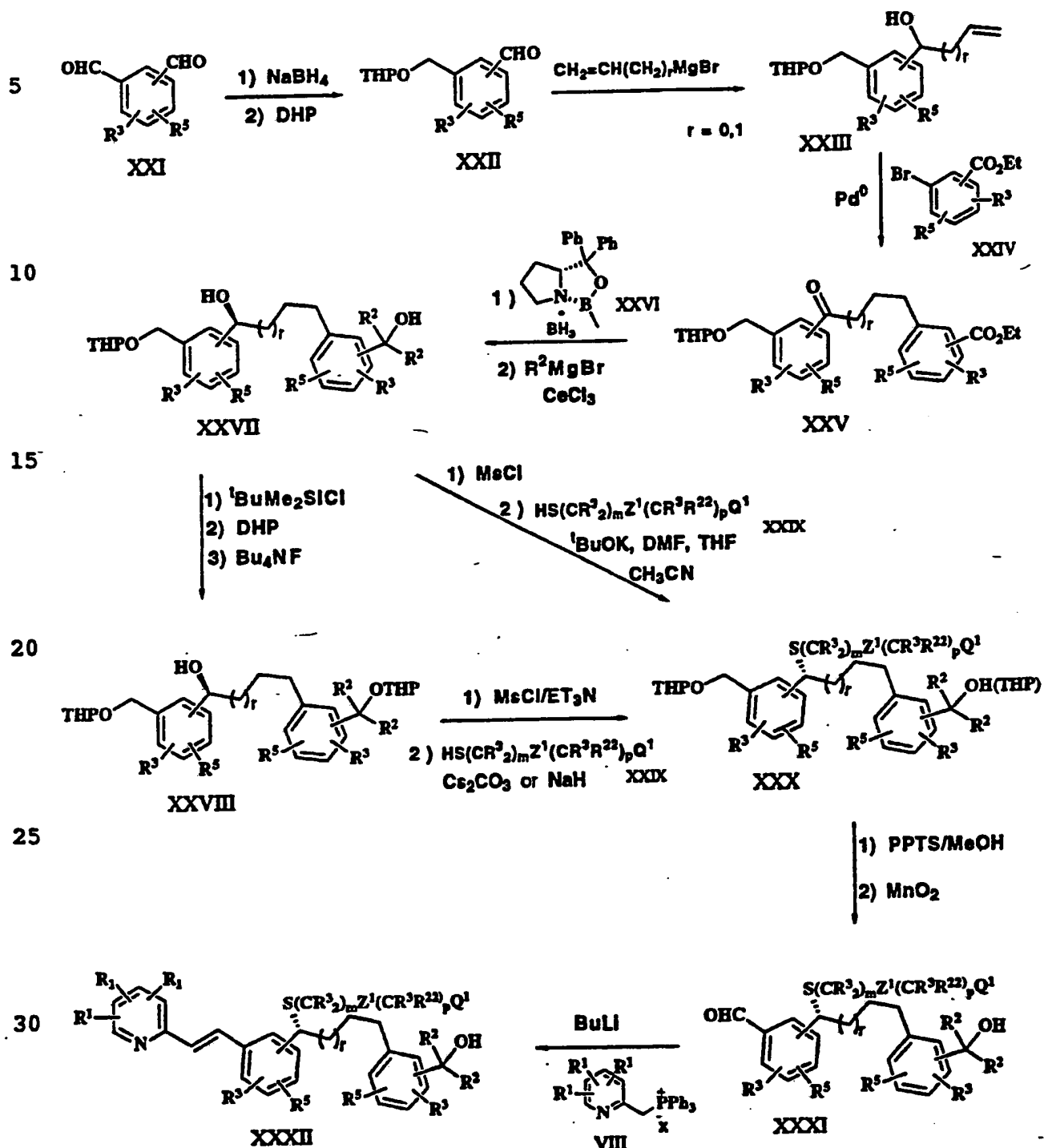
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## SCHEME 4



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## SCHEME 5

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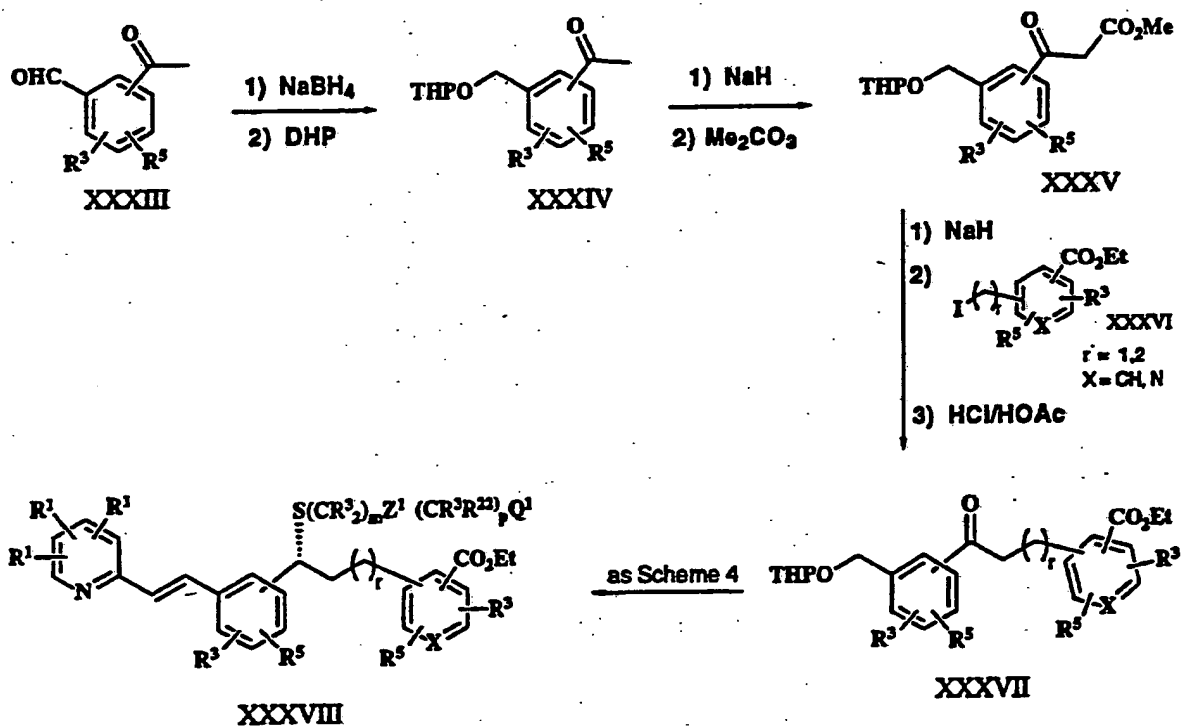
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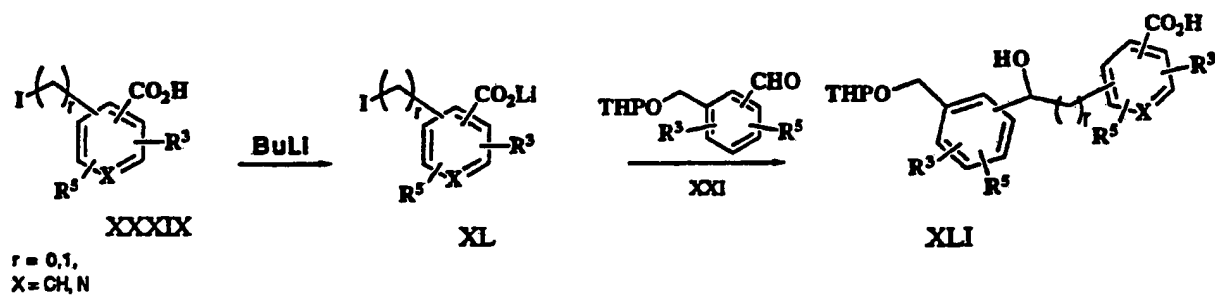
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## SCHEME 6

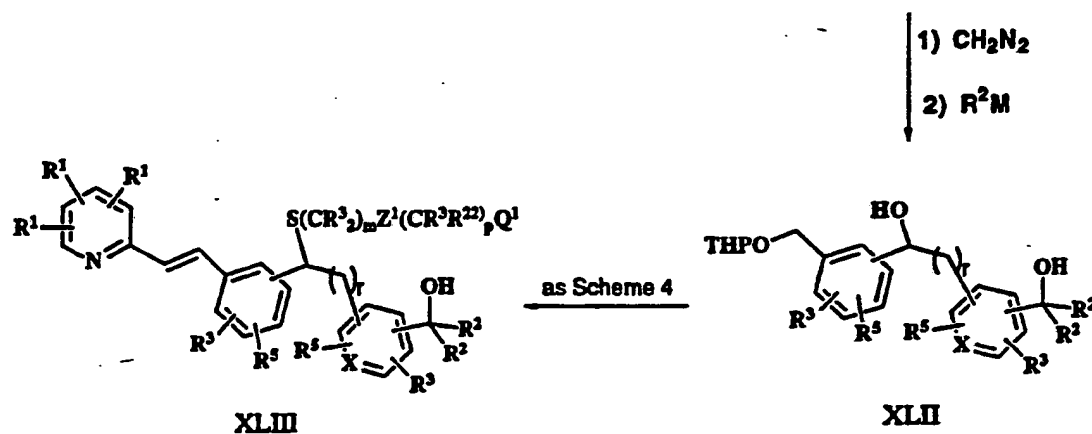
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SCHEME 7

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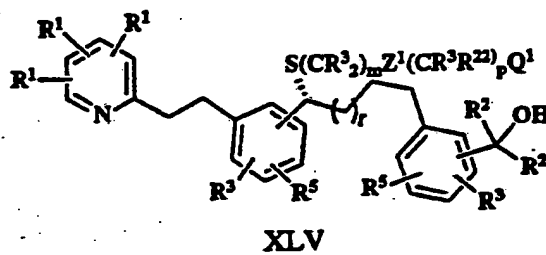
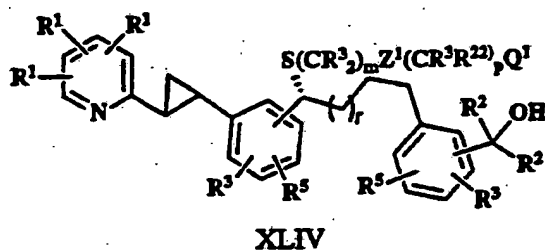
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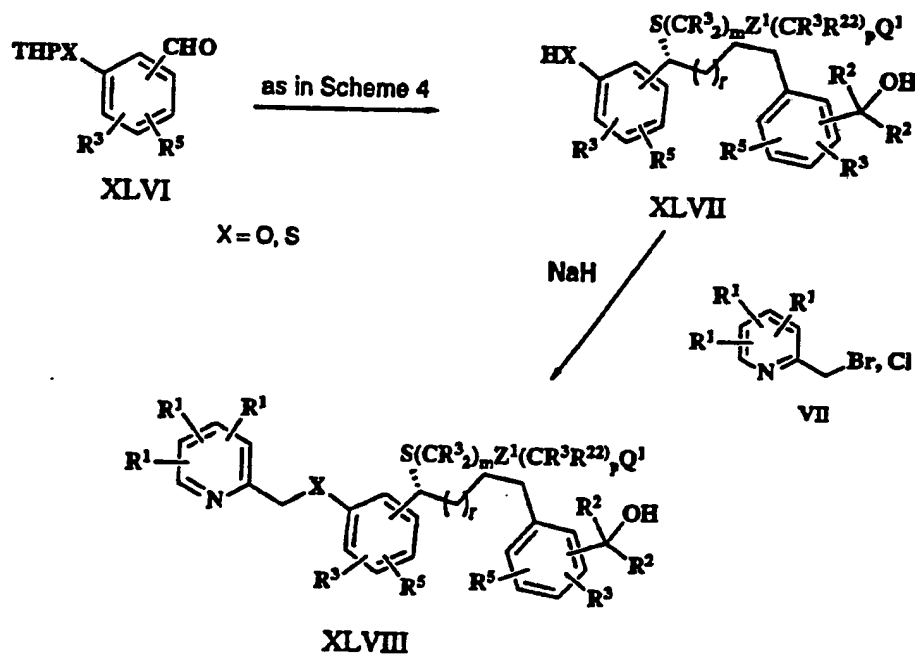
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XXXII

 $(\text{CH}_3)_3\text{S}^+\text{I}^-$  $\text{BH}_3 \cdot \text{THF}$ 

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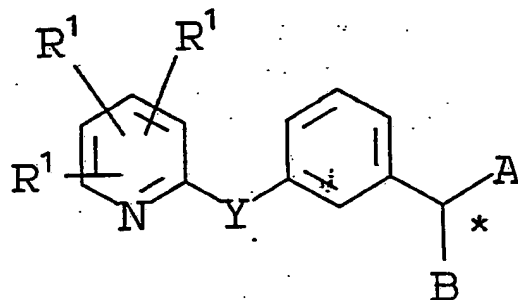
SCHEME 8

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REPRESENTATIVE COMPOUNDS

Table I illustrates compounds of formula Ib, which are representative of the present invention.



1b

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TABLE I

EX. *	R <sup>1</sup>	R <sup>1</sup>	Y	A	B
1	R	5-Me	6-Me	(CH <sub>2</sub> ) <sub>2</sub> (1,2-phe)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>2</sub> CH <sub>2</sub> COOH
2	R	5-CF <sub>3</sub>	H	(CH <sub>2</sub> ) <sub>2</sub> (1,2-phe)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>2</sub> CH <sub>2</sub> COOH
3	R	6-Cl	H	(CH <sub>2</sub> ) <sub>2</sub> (1,2-phe)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>2</sub> CH <sub>2</sub> COOH
4	R	5-n-Bu	H	(CH <sub>2</sub> ) <sub>2</sub> (1,2-phe)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>2</sub> CH <sub>2</sub> COOH
5	R	4-Me	6-Me	(CH <sub>2</sub> ) <sub>2</sub> (1,2-phe)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>2</sub> CH <sub>2</sub> COOH
6	R	H	H	(CH <sub>2</sub> ) <sub>2</sub> (1,2-phe)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>2</sub> CH <sub>2</sub> COOH
7	R	5-OMe	H	(CH <sub>2</sub> ) <sub>2</sub> (1,2-phe)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>2</sub> CH <sub>2</sub> COOH
8	R	5,6-(CH <sub>2</sub> ) <sub>4</sub> -	-	(CH <sub>2</sub> ) <sub>2</sub> (1,2-phe)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>2</sub> CH <sub>2</sub> COOH
9	R	5,6-(CH <sub>2</sub> ) <sub>3</sub> -	-	(CH <sub>2</sub> ) <sub>2</sub> (1,2-phe)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>2</sub> CH <sub>2</sub> COOH
10	R	5-ph	H	(CH <sub>2</sub> ) <sub>2</sub> (1,2-phe)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>2</sub> CH <sub>2</sub> COOH
11	R	6-i-Pr	H	(CH <sub>2</sub> ) <sub>2</sub> (1,2-phe)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>2</sub> CH <sub>2</sub> COOH
12	R	5-Et	6-Me	(CH <sub>2</sub> ) <sub>2</sub> (1,2-phe)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>2</sub> CH <sub>2</sub> COOH
13	R	6-n-Bu	H	(CH <sub>2</sub> ) <sub>2</sub> (1,2-phe)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>2</sub> CH <sub>2</sub> COOH
14	R	6-c-Pen	H	(CH <sub>2</sub> ) <sub>2</sub> (1,2-phe)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>2</sub> CH <sub>2</sub> COOH
15	R	5-CF <sub>3</sub>	6-c-Bu	(CH <sub>2</sub> ) <sub>2</sub> (1,2-phe)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>2</sub> CH <sub>2</sub> COOH
16	R	5-F	6-i-Bu	(CH <sub>2</sub> ) <sub>2</sub> (1,2-phe)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>2</sub> CH <sub>2</sub> COOH
17	R	6-(4-F-Ph)	H	(CH <sub>2</sub> ) <sub>2</sub> (1,2-phe)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>2</sub> CH <sub>2</sub> COOH
18	R	5-CF <sub>3</sub> CH <sub>2</sub>	6-c-Pr-CH <sub>2</sub>	(CH <sub>2</sub> ) <sub>2</sub> (1,2-phe)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>2</sub> CH <sub>2</sub> COOH
19	R	5-Cl	6-Ac	(CH <sub>2</sub> ) <sub>2</sub> (1,2-phe)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> CH(CH <sub>3</sub> )CH <sub>2</sub> Tz

TABLE I (cont)

EX. *	R <sup>1</sup>	R <sup>1</sup>	Y	A	B
20	RS	5,6-CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> CHMe-	CH=CH	(CH <sub>2</sub> ) <sub>2</sub> ((4-F-1,2-phe)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> CH(CH <sub>3</sub> )CH <sub>2</sub> Tz
21	R	5,6-(CH <sub>2</sub> ) <sub>6</sub> -	CH <sub>2</sub> O	(CH <sub>2</sub> ) <sub>2</sub> (4-Cl-1,2-phe)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>2</sub> CH <sub>2</sub> COOH
22	RS	6-(2-Thia) H	CH <sub>2</sub> CH <sub>2</sub>	(CH <sub>2</sub> ) <sub>2</sub> (3-CF <sub>3</sub> -1,2-phe)CHMeOH	SCH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> COOH
23	S	5-Et	CH <sub>2</sub> S <sup>1</sup>	CH <sub>2</sub> (2,3-Thio)C(CH <sub>2</sub> ) <sub>2</sub> OH	SCH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> COOH
24	R	6-(2-Fu)	CH=CH	CH <sub>2</sub> (2,3-fur)CETOH	SCH <sub>2</sub> C(Me) <sub>2</sub> CONHSO <sub>2</sub> CF <sub>3</sub>
25	S	6-Th	CH <sub>2</sub> O	(CH <sub>2</sub> ) <sub>2</sub> (1,2-phe)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>2</sub> CH <sub>2</sub> CO <sub>2</sub> H
26	R	5-CF <sub>3</sub> CH <sub>2</sub> CH <sub>2</sub>	CH=CH	(CH <sub>2</sub> ) <sub>2</sub> (4-Cl-1,2-phe)C(Me) <sub>2</sub> OH	OCH <sub>2</sub> C(CH <sub>3</sub> ) <sub>2</sub> CH <sub>2</sub> CO <sub>2</sub> H
27	RS	5-i-Pr	CH=CH	(CH <sub>2</sub> ) <sub>2</sub> (4-Cl-1,2-phe)C(Me) <sub>2</sub> OH	OCH <sub>2</sub> CH(CH <sub>3</sub> )CH <sub>2</sub> Tz
28	RS	3-F	CH <sub>2</sub> O	(CH <sub>2</sub> ) <sub>2</sub> (4-F-1,2-phe)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> CH(C <sub>2</sub> H <sub>5</sub> )CH <sub>2</sub> CONME <sub>2</sub>
29	S	5-c-Pr	CH <sub>2</sub> CH <sub>2</sub>	CH <sub>2</sub> (4-F-1,2-phe)CMePhOH	SCH <sub>2</sub> CH <sub>2</sub> CO <sub>2</sub> H
30	R	5,6-(CH <sub>2</sub> ) <sub>4</sub> -	CH <sub>2</sub> CH <sub>2</sub>	(CH <sub>2</sub> ) <sub>2</sub> (3-Cl-1,2-phe)CHMeOH	SCH <sub>2</sub> CH <sub>2</sub> CONHS(O) <sub>2</sub> Ph
31	R	5,6-CH <sub>2</sub> CH <sub>2</sub> CH(Me)-	CH <sub>2</sub> S	(CH <sub>2</sub> ) <sub>2</sub> (5-F-1,2-phe)CMeCF <sub>3</sub> OH	SCH <sub>2</sub> CH <sub>2</sub> CONHS(O) <sub>2</sub> CH <sub>3</sub>
32	RS	5-MeOCH <sub>2</sub>	CH=CH	(6-CF <sub>3</sub> -1,2-phe)CHCF <sub>3</sub> OH	SCH <sub>2</sub> CH <sub>2</sub> CONHS(O) <sub>2</sub> CF <sub>3</sub>
33	R	5-Me	CH <sub>2</sub> O	(CH <sub>2</sub> ) <sub>2</sub> (4-CF <sub>3</sub> -1,2-phe)C(CF <sub>3</sub> ) <sub>2</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>2</sub> CH <sub>2</sub> CONHS(O) <sub>2</sub> Ph
34	R	5-i-C <sub>3</sub> H <sub>5</sub>	CH=CH	(CH <sub>2</sub> ) <sub>2</sub> (4-F-1,3-phe)CMeEtOH	SCH <sub>2</sub> CH <sub>2</sub> C(CH <sub>3</sub> ) <sub>2</sub> CO <sub>2</sub> H
35	RS	6-(2-Py)	CH=CH	(CH <sub>2</sub> ) <sub>2</sub> (4-F-1,4-phe)C(CH <sub>2</sub> ) <sub>2</sub> OH	SCH <sub>2</sub> CH(C <sub>2</sub> H <sub>5</sub> )Tz
36	S	5-Cl	CH=CH	(CH <sub>2</sub> ) <sub>2</sub> (4-F-1,2-phe)C(CH <sub>2</sub> ) <sub>3</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>2</sub> NHS(O) <sub>2</sub> CF <sub>3</sub>
37	R	4-F	CH <sub>2</sub> S	(CH <sub>2</sub> ) <sub>2</sub> (4-F-1,2-phe)C(CH <sub>2</sub> ) <sub>4</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>3</sub> CH <sub>2</sub> CO <sub>2</sub> H
38	R	3-MeO	CH=CH	(CH <sub>2</sub> ) <sub>2</sub> (4-F-1,2-phe)C(CH <sub>2</sub> ) <sub>5</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>4</sub> CH <sub>2</sub> CO <sub>2</sub> H
39	R	5-F	CH <sub>2</sub> CH <sub>2</sub>	(CH <sub>2</sub> ) <sub>2</sub> (2,5-fur)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>5</sub> CH <sub>2</sub> CO <sub>2</sub> H
40	RS	6-(2-Ox)	CH=CH	(CH <sub>2</sub> ) <sub>2</sub> (1,2-phe)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> C(CH <sub>3</sub> ) <sub>2</sub> CH <sub>2</sub> CO <sub>2</sub> H



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Assays For Determining Biological Activity

The leukotriene antagonist properties of the compounds of the present invention are evaluated using the following assays.

5

Three assays are described in T.R. Jones et al., Can. J. Physiol. Pharmacol., 1989, 67, 17-28.

These are:

10

- 1) LTD<sub>4</sub> Receptor Binding Assays in Guinea Pig Lung Membranes,
- 2) Guinea Pig Trachea, and
- 3) In Vivo Assays in Anesthetized Guinea Pigs.

Asthmatic Rat Assay

15

Rats are obtained from an inbred line of asthmatic rats. Both female (190-250 g) and male (260-400 g) rats are used.

20

Egg albumin (EA), grade V, crystallized and lyophilized, is obtained from Sigma Chemical Co., St. Louis. Aluminum hydroxide is obtained from the Regis Chemical Company, Chicago. Methysergide bimaleate is supplied by Sandoz Ltd., Basel.

25

The challenge and subsequent respiratory recordings are carried out in a clear plastic box with internal dimensions 10x6x4 inches. The top of the box is removable; in use, it is held firmly in place by four clamps and an airtight seal is maintained by a soft rubber gasket. Through the center of each end of the chamber a DeVilbiss nebulizer (No. 40) is inserted via an airtight seal

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and each end of the box also has an outlet. A Fleisch No. 0000 pneumotachograph is inserted into one end of the box and coupled to a Grass volumetric pressure transducer (PT5-A) which is then connected to a Beckman Type R Dynograph through appropriate couplers. While aerosolizing the antigen, the outlets are open and the pneumotachograph is isolated from the chamber. The outlets are closed and the pneumotachograph and the chamber are connected during the recording of the respiratory patterns. For challenge, 2 mL of a 3% solution of antigen in saline is placed into each nebulizer and the aerosol is generated with air from a small Potter diaphragm pump operating at 10 psi and a flow of 8 liters/minute.

Rats are sensitized by injecting (subcutaneously) 1 mL of a suspension containing 1 mg EA and 200 mg aluminum hydroxide in saline. They are used between days 12 and 24 postsensitization. In order to eliminate the serotonin component of the response, rats are pretreated intravenously 5 minutes prior to aerosol challenge with 3.0 mg/kg of methysergide. Rats are then exposed to an aerosol of 3% EA in saline for exactly 1 minute, then their respiratory profiles are recorded for a further 30 minutes. The duration of continuous dyspnea is measured from the respiratory recordings.

Compounds are generally administered either orally 1-4 hours prior to challenge or intravenously 2 minutes prior to challenge. They are either dissolved in saline or 1% methocel or suspended in 1% methocel. The volume injected is 1 mL/kg (intravenously) or 10 mL/kg (orally). Prior to oral treatment rats are starved overnight. Their activity

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is determined in terms of their ability to decrease the duration of symptoms of dyspnea in comparison with a group of vehicle-treated controls. Usually, a compound is evaluated at a series of doses and an  
5 ~~ED<sub>50</sub> is determined. This is defined as the dose~~  
(mg/kg) which would inhibit the duration of symptoms by 50%.

10 Pulmonary Mechanics In Trained Conscious Squirrel Monkeys

The test procedure involves placing trained squirrel monkeys in chairs in aerosol exposure chambers. For control purposes, pulmonary mechanics measurements of respiratory parameters are recorded  
15 for a period of about 30 minutes to establish each monkey's normal control values for that day. For oral administration, compounds are dissolved or suspended in a 1% methocel solution (methylcellulose, 65HG, 400 cps) and given in a volume of 1 mL/kg body  
20 weight. For aerosol administration of compounds, a DeVilbiss ultrasonic nebulizer is utilized. Pretreatment periods vary from 5 minutes to 4 hours before the monkeys are challenged with aerosol doses of either leukotriene D<sub>4</sub> (LTD<sub>4</sub>) or Ascaris suum  
25 antigen.

Following challenge, each minute of data is calculated by computer as a percent change from control values for each respiratory parameter  
including airway resistance ( $R_L$ ) and dynamic  
30 compliance ( $C_{dyn}$ ). The results for each test compound are subsequently obtained for a minimum period of 60 minutes post challenge which are then compared to previously obtained historical baseline

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control values for that monkey. In addition, the overall values for 60 minutes post-challenge for each monkey (historical baseline values and test values) are averaged separately and are used to calculate the overall percent inhibition of LTD<sub>4</sub> or Ascaris antigen response by the test compound. For statistical analysis, paired t-test is used. (References: McFarlane, C.S. et al., Prostaglandins, 28, 173-182 (1984) and McFarlane, C.S. et al., Agents Actions, 22, 63-68 (1987).)

#### Prevention Of Induced Bronchoconstriction In Allergic Sheep

##### 15           A. Rationale:

Certain allergic sheep with known sensitivity to a specific antigen (Ascaris suum) respond to inhalation challenge with acute and late bronchial responses. The time course of both the acute and the late bronchial responses approximates the time course observed in asthmatics and the pharmacological modification of both responses is similar to that found in man. The effects of antigen in these sheep are largely observed in the large airways and are conveniently monitored as changes in lung resistance or specific lung resistance.

##### B. Methods:

Animal Preparation: Adult sheep with a mean weight of 35 kg (range, 18 to 50 kg) are used. All animals used meet two criteria: a) they have a natural cutaneous reaction to 1:1,000 or 1:10,000

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dilutions of Ascaris suum extract (Greer Diagnostics, Lenois, NC) and b) they have previously responded to inhalation challenge with Ascaris suum with both an acute bronchoconstriction and a late bronchial obstruction (W.M. Abraham et al., Am. Rev. Resp. Dis., 128, 839-44 (1983)).

Measurement of Airway Mechanics: The unsedated sheep are restrained in a cart in the prone position with their heads immobilized. After topical anesthesia of the nasal passages with 2% lidocaine solution, a balloon catheter is advanced through one nostril into the lower esophagus. The animals are then intubated with a cuffed endotracheal tube through the other nostril using a flexible fiberoptic bronchoscope as a guide. Pleural pressure is estimated with the esophageal balloon catheter (filled with one ml of air), which is positioned such that inspiration produces a negative pressure deflection with clearly discernible cardiogenic oscillations. Lateral pressure in the trachea is measured with a sidehole catheter (inner dimension, 2.5 mm) advanced through and positioned distal to the tip of the nasotracheal tube. Transpulmonary pressure, the difference between tracheal pressure and pleural pressure, is measured with a differential pressure transducer (DP45; Validyne Corp., Northridge, CA). For the measurement of pulmonary resistance ( $R_L$ ), the maximal end of the nasotracheal tube is connected to a pneumotachograph (Fleisch, Dyna Sciences, Blue Bell, PA). The signals of flow and transpulmonary pressure are recorded on an oscilloscope (Model DR-12; Electronics for Medicine,

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White Plains, NY) which is linked to a PDP-11 Digital computer (Digital Equipment Corp., Maynard, MA) for on-line calculation of  $R_L$  from transpulmonary pressure, respiratory volume obtained by integration and flow. Analysis of 10-15 breaths is used for the determination of  $R_L$ . Thoracic gas volume ( $V_{tg}$ ) is measured in a body plethysmograph, to obtain specific pulmonary resistance ( $SR_L = R_L \cdot V_{tg}$ ).

Aerosol Delivery Systems: Aerosols of Ascaris suum extract (1:20) are generated using a disposable medical nebulizer (Raindrop®, Puritan Bennett), which produces an aerosol with a mass median aerodynamic diameter of 6.2  $\mu$ M (geometric standard deviation, 2.1) as determined by an electric size analyzer (Model 3030; Thermal Systems, St. Paul, MN). The output from the nebulizer is directed into a plastic t-piece, one end of which is attached to the nasotracheal tube, the other end of which is connected to the inspiratory part of a Harvard respirator. The aerosol is delivered at a tidal volume of 500 mL of a rate of 20 per minute. Thus, each sheep receives an equivalent dose of antigen in both placebo and drug trials.

Experimental Protocol: Prior to antigen challenge baseline measurements of  $SR_L$  are obtained, infusion of the test compound is started 1 hr prior to challenge, the measurement of  $SR_L$  repeated and then the sheep undergoes inhalation challenge with Ascaris suum antigen. Measurements of  $SR_L$  are obtained immediately after antigen challenge and at 1, 2, 3, 4, 5, 6, 6.5, 7, 7.5, and 8 hrs after antigen challenge. Placebo and drug tests are separated by at least 14 days. In a further study,

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sheep are given a bolus dose of the test compound followed by an infusion of the test compound for 0.5-1 hr prior to Ascaris challenge and for 8 hrs after Ascaris as described above.

5                    Statistical Analysis: A Kruskal-Wallis one way ANOVA test is used to compare the acute immediate responses to antigen and the peak late response in the controls and the drug treated animals.

10                   The invention is further defined by the following non-limiting examples in which, unless stated otherwise:

15                   (i) all operations were carried out under a nitrogen atmosphere at room or ambient temperature, that is, at a temperature in the range 18-25°C;

20                   (ii) evaporation of solvent was carried out using a rotary evaporator under reduced pressure (600-4000 pascals: 4.5-30 mm. Hg) with a bath temperature of up to 60°C;

25                   (iii) the course of reactions was followed by thin layer chromatography (TLC) and reaction times are given for illustration only;

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- 5 (iv) melting points are uncorrected and 'd' indicates decomposition; the melting points given are those obtained for the materials prepared as described; polymorphism may result in isolation of materials with different melting points in some preparations;
- 10 (v) final products were essentially pure by TLC and had satisfactory nuclear magnetic resonance (NMR) spectra, microanalytical data, and/or mass spectra;
- 15 (vi) yields are given for illustration only and, for crystalline end-products, refer to the weight of recrystallized solid;
- 20 (vii) when given, NMR data are in the form of delta ( $\delta$ ) values for major diagnostic protons, given in parts per million (ppm) relative to tetramethylsilane (TMS) as internal standard, determined at 250 MHz or 300 MHz using the indicated solvent;
- 25 conventional abbreviations for signal shape are used for example, s. singlet; d. doublet; m. multiplet; br. broad); "Ar" signifies an aromatic signal;
- 30 (viii) chemical symbols have their usual meanings; the following abbreviations have also been used: v (volume), w (weight), b.p. (boiling point), m.p. (melting point), L (liter(s)), mL (milliliters),  $\mu$ L (microliters), g

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(gram(s)), mg (milligram(s)), mol (moles),  
mmol (millimoles), eq. (equivalent(s)), h  
(hour(s)).

5

EXAMPLE 1

(R)-Sodium 1-(((1-(3-(2-(5,6-dimethyl-2-pyridinyl)  
ethenyl)phenyl)-3-(2-(2-hydroxy-2-propyl)phenyl)  
propyl)thio)methyl)cyclopropaneacetate

10

Step 1: 2,3-Dimethylpyridine N-oxide

To the 2,3-dimethylpyridine (15.0 g, 140  
mmol) in  $\text{CHCl}_3$  (40 mL) at 0°C was slowly added a  
15  $\text{CHCl}_3$  (200 mL) solution of m-chloroperbenzoic acid  
(21.5 g, 154 mmol). The mixture was stirred 1 hour  
at 0°C, the ice bath was removed and stirring  
continued another hour. Calcium hydroxide (26 g,  
350 mmol) was added and the slurry was vigorously  
20 stirred 4.5 hours before filtering through celite.  
The cake was thoroughly washed with  $\text{CH}_2\text{Cl}_2$ .  
Evaporation of the solvents left an oily solid that  
was swished in  $\text{Et}_2\text{O}$  giving 8.71 g of desired product  
which was used without further purification.

25

Step 2: 5,6-Dimethyl-2-pyridinecarbonitrile

To a slurry of N-oxide from Step 1 (8.7 g)  
in  $\text{CH}_2\text{Cl}_2$  (125 mL) at r.t. was added trimethyl-  
30 silylcyanide (9.9 mL, 74.2 mmol). After stirring for  
15 minutes, N,N-diethylcarbamoyl chloride (9.4 mL,  
74.2 mmol) was added and the resulting mixture was

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allowed to stir at r.t. for 2.5 days. The reaction was quenched by careful addition of 10% aq.  $K_2CO_3$ , stirred 15 minutes, and extracted (3x) with  $CH_2Cl_2$ . The organic layer was washed with aq.  $K_2CO_3$ , brine, and dried over  $Na_2SO_4/K_2CO_3$ . Evaporation of the solvent and purification of the residue by flash chromatography (25% to 50%  $Et_2O$  in hexanes) yielded 4.13 g of the title compound.

$^1H$  NMR ( $CD_3COCD_3$ ):  $\delta$  7.73 (1H, d), 7.62 (1H, d), 2.51 (3H, s), 2.39 (3H, s).

Step 3: Methyl 5,6-dimethyl-2-pyridinecarboxylate

Dry HCl (gas) was bubbled through anhydrous MeOH (125 mL) at  $-10^\circ C$  until saturation. The cyanopyridine from Step 2 (1.33 g, 10 mmol) was added and HCl was bubbled for another 5 minutes. The flask was sealed and the reaction allowed to stir 3 days at r.t. After carefully depressurizing the flask, water (3 mL) was added before concentrating in vacuo. The residue was diluted with EtOAc and saturated aq.  $NaHCO_3$ . Extraction of the aqueous phase with EtOAc (2x) followed by washing of the organic layers with aq.  $NaHCO_3$ , and brine gave upon concentration in vacuo 1.59 g (96%) of desired carbomethoxypyridine.

$^1H$  NMR ( $CD_3COCD_3$ ):  $\delta$  7.8 (1H, d), 7.65 (1H, d), 3.85 (3H, s), 2.5 (3H, s), 2.35 (3H, s).

Step 4: 5,6-Dimethyl-2-(hydroxymethyl)pyridine

Diisobutyl aluminum hydride (4.45 mL, 25 mmol) was added to the carbomethoxypyridine from Step 3 (1.4 g, 8.4 mmol) in THF (40 mL) at  $-78^\circ C$ . The

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mixture was stirred overnight with slow warming to +4°C. The reaction was quenched with solid tartaric acid followed by aqueous sodium potassium tartrate and stirred 1/2 hour. Neutralization with saturated  
5 aq. NaHCO<sub>3</sub> and extraction with EtOAc (3x) gave after treatment with brine and evaporation 1.12 g (97%) of the title compound.

<sup>1</sup>H NMR (CD<sub>3</sub>COCD<sub>3</sub>): δ 7.5 (1H, d), 7.15 (1H, d), 4.45 (2H, brs), 2.4 (3H, s), 2.22 (3H, s).

10

Step 5: 2-Bromomethyl-5,6-dimethylpyridine

Bromine (1M/CCl<sub>4</sub>, 3.6 mL, 3.6 mmol) was added to a -5°C solution of triphenylphosphine (943  
15 mg, 3.6 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (12 mL). The color faded away and a solid precipitated. The mixture was warmed to r.t. and the pyridinecarbinol from Step 4 (411 mg, 3.0 mmol) was added dropwise as a CH<sub>2</sub>Cl<sub>2</sub> (5 mL) solution. After stirring 1 hour, the reaction  
20 mixture was quenched with saturated aq. NaHCO<sub>3</sub>, extracted with EtOAc (3x) and the combined organic extracts were washed with brine and dried over Na<sub>2</sub>SO<sub>4</sub>. Evaporation of the solvents left a crude residue (1.8 g) which was purified on a short silica  
25 gel column (20% EtOAc in hexanes) to give 546 mg (91%) of crystals of the title compound.

<sup>1</sup>H NMR (CD<sub>3</sub>COCD<sub>3</sub>): δ 7.5 (1H, d), 7.25 (1H, d), 4.45 (2H, s), 2.42 (3H, s), 2.27 (3H, s).

30 Step 6: ((5,6-Dimethyl-2-pyridinyl)methyl)triphenylphosphonium bromide

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The bromide from Step 5 (740 mg, 3.7 mmol) and triphenylphosphine (1.31 g, 5 mmol) were dissolved in acetonitrile (15 mL) and the mixture was refluxed for 6 hours. Evaporation of the solvent in vacuo left a solid that was swished in Et<sub>2</sub>O. Filtration and vacuum drying of the powder gave 1.46 g (86%) of the desired phosphonium salt.

10 Step 7: 1,1-Cyclopropanedimethanol cyclic sulfite

To a solution of BH<sub>3</sub>:THF complex (1M in THF, 262 mL) was added diethyl 1,1-cyclopropane-dicarboxylate (25 g, 134 mmol) at 25°C under N<sub>2</sub>. The solution was heated at reflux for 6 hr, cooled to r.t., and MeOH (300 mL) was cautiously added. The solution was stirred for 1 hr and then concentrated to an oil. The crude diol was dissolved in CH<sub>2</sub>Cl<sub>2</sub> (234 mL) and SOCl<sub>2</sub> (15.9 g, 134 mmol) was added dropwise over a period of 15 min at 25°C. After stirring for another 15 min, the mixture was washed with aqueous NaHCO<sub>3</sub>. The organic extract was dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and concentrated to give quantitatively the title compound as a white solid.

25 Step 8: 1-(Hydroxymethyl)cyclopropaneacetonitrile

To a solution of the cyclic sulfite product of step 7 (14.7 g, 99 mmol) in DMF (83 mL) was added NaCN (9.74 g, 199 mmol). The mixture was heated to 90°C for 20 h. Upon cooling, EtOAc (400 mL) was added and the solution was washed with saturated NaHCO<sub>3</sub> solution (55 mL), H<sub>2</sub>O (4 x 55mL), saturated NaCl solution and dried over Na<sub>2</sub>SO<sub>4</sub>. The solution was concentrated to give 7.1 g (65%) of the title compound.

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Step 9: 1-(Acetythiomethyl)cyclopropaneacetonitrile

To a solution of the alcohol of Step 8 (42 g, 378 mmol) in dry  $\text{CH}_2\text{Cl}_2$  (450 mL) at  $-30^\circ\text{C}$  was added  $\text{Et}_3\text{N}$  (103.7 mL, 741 mmol) followed by  $\text{CH}_3\text{SO}_2\text{Cl}$  (43.3 mL, 562 mmol) dropwise. The mixture was warmed to  $25^\circ\text{C}$ , washed with  $\text{NaHCO}_3$ , dried over  $\text{Na}_2\text{SO}_4$  and concentrated in vacuo to give the corresponding mesylate. The mesylate was then dissolved in DMF (450 mL) and cooled to  $0^\circ\text{C}$ . Potassium thioacetate (55.4 g, 485 mmol) was added, and the mixture was stirred at  $25^\circ\text{C}$  for 18 hr.  $\text{EtOAc}$  (1.5 L) was added, the solution was washed with  $\text{NaHCO}_3$ , dried over  $\text{Na}_2\text{SO}_4$  and concentrated in vacuo to give 45 g (70%) of the title compound.

Step 10: Methyl 1-(thiomethyl)cyclopropaneacetate

To a solution of the nitrile of Step 9 (45 g, 266 mmol) in MeOH (1.36 L) was added  $\text{H}_2\text{O}$  (84 mL) and conc.  $\text{H}_2\text{SO}_4$  (168 mL). The mixture was heated to reflux for 20 hr, cooled to  $25^\circ\text{C}$ ,  $\text{H}_2\text{O}$  (1 L) was added and the product was extracted with  $\text{CH}_2\text{Cl}_2$  (2x 1.5 L). The organic extract was washed with  $\text{H}_2\text{O}$  and dried over  $\text{Na}_2\text{SO}_4$ . Concentration of the organic solution gave 36 g (93%) of the title compound.

Step 11: 3-((2-Tetrahydropyranyl)oxymethyl) benzaldehyde

Isophthalaldehyde (150 g, 1.1 mole) was dissolved in THF (1 L) and EtOH (1 L) at  $0^\circ\text{C}$ . Sodium borohydride (11.0 g, 291 mmol) was added portionwise

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and the mixture stirred 1 hour at 0°C. Addition of 25% aq.  $\text{NH}_4\text{OAc}$  and extraction with  $\text{EtOAc}$  (2x) gave after evaporation the crude product. Purification by flash chromatography (20% to 40%  $\text{EtOAc}$  in hexanes) yielded 60 g of m-hydroxymethylbenzaldehyde.

This alcohol (0.44 mole) was dissolved in  $\text{CH}_2\text{Cl}_2$  (500 mL), dihydropyran (50 g, 0.59 mole) and p-toluenesulfonic acid (1g, 5 mmol) were added and the mixture was stirred overnight at r.t. After concentration in vacuo, the residue was purified by flash chromatography (5% to 15%  $\text{EtOAc}$  in toluene) to give 85 g of the title compound.

Step 12:      3-((2-Tetrahydropyranyl)oxymethyl)-2-propene-1-ol

To the aldehyde of Step 11 (85 g, 386 mmol) in toluene (1 L) at 0°C was slowly added vinyl magnesium bromide (450 mL, 1M, 450 mmol) over a 30 minute period. After stirring for 1 hour the reaction mixture was quenched with 25% aq.  $\text{NH}_4\text{OAc}$  and extracted with  $\text{EtOAc}$  (3x). Evaporation and purification by flash chromatography (15% to 25%  $\text{EtOAc}$  in toluene) yielded 82 g (86%) of the title compound.

Step 13:      Ethyl 2-(3-(3-((2-tetrahydropyranyl)oxymethyl)phenyl)-3-oxopropyl)benzoate

The allylic alcohol of Step 12 (24.8 g, 100 mmol) and ethyl o-bromobenzoate (25.2 g, 110 mmol) were dissolved in DMF (200 mL). Lithium chloride

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(4.2 g, 100 mmol), lithium acetate dihydrate (25.5 g, 250 mmol), and tetra-n-butylammonium chloride (55 g, 200 mmol) were added and the resulting mixture was degassed three times. Palladium (II) acetate (1 g) was then added and the mixture was degassed three more times before heating it at 100°C while stirring for 1 hour. After cooling to r.t., the reaction mixture was poured onto H<sub>2</sub>O (600 mL), 10% aq. NaHCO<sub>3</sub> (200 mL) and Et<sub>2</sub>O. The crude product was extracted with Et<sub>2</sub>O (2x), washed with H<sub>2</sub>O and brine, and dried over Na<sub>2</sub>SO<sub>4</sub> before concentrating in vacuo. Purification on a short silica gel column (20% EtOAc in hexanes) gave 34 g (86%) of the title compound. <sup>1</sup>H NMR (CD<sub>3</sub>COCD<sub>3</sub>): δ 8.02 (1H, brs), 7.92 (1H, d), 7.88 (1H, d), 7.62 (1H, d), 7.50 (3H, m), 7.32 (1H, brt), 4.8 (1H, d), 4.70 (1H, brs), 4.54 (1H, d), 4.3 (2H, q), 3.82 (1H, m), 3.50 (1H, m), 3.35 (4H, m), 1.9-1.45 (6H, m), 1.32 (3H, t).

20 Step 14: Ethyl 2-(3-(3-((2-tetrahydropyranyl)-oxymethyl)phenyl)-3(S)-hydroxypropyl)-benzoate

The keto ester of Step 13 (24.8 g, 62.5 mmol) was dissolved in THF (230 mL) and cooled to -45°C. A THF (15 mL) solution of tetrahydro-1-methyl-3,3-diphenyl-1H,3H-pyrrolo[1,2-c][1,3,2]-oxazaborole-borane adduct (J. Org. Chem., 56, 751 (1991), 4.55 g, 15.6 mmol) was added dropwise and the resulting mixture was stirred 20 minutes at -45°C. To this solution, 1.0M borane-THF (62.5 mL, 62.5 mmol) was added dropwise over 30 minutes. The

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reaction mixture was stirred 1 hour at -45°C followed by another 2 hours with slow warming to -20°C. After cooling the solution of -40°C, it was poured onto 25% aq. NH<sub>4</sub>OAc (425 mL) and 1.0M diethanolamine (40 mL) at 0°C and stirred vigorously for 20 minutes. The title compound was extracted with EtOAc (3x), dried over MgSO<sub>4</sub> and concentrated under reduced pressure. The crude oil was purified by flash chromatography (25% to 50% EtOAc in hexanes) to yield 22.6 g (91%) of the desired product as an oil.  
[α]<sub>D</sub><sup>25</sup> = -32.6° (C = 3, CHCl<sub>3</sub>)

Step 15:        2-(3-(3-((2-Tetrahydropyranyl)oxymethyl)phenyl)-3(S)-hydroxypropyl)-α,α-dimethylbenzenemethanol

Anhydrous CeCl<sub>3</sub> (17.25 g, 70 mmol) was refluxed for 2.5 hours in THF (200 mL) using a Dean-Stark trap filled with molecular sieves to remove H<sub>2</sub>O. The ivory suspension was cooled to -5°C and MeMgCl (114 mL, 3M/THF, 340 mmol) was added dropwise while keeping the internal temperature between -10°C and 0°C. The grey suspension was stirred 2 hours before slowly adding to it the hydroxy-ester of Step 14 (27.1 g, 68 mmol) as a THF solution (200 mL) via a cannula. The resulting mixture was stirred 1.5 hours at or below 0°C, and then slowly poured onto ice cold 1M AcOH (1 L) and EtOAc (500 mL) and stirred for 30 minutes. After adjusting the pH to 6-7, the crude compound was extracted with EtOAc (2x) and the combined organic

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phases were washed with saturated aq.  $\text{NaHCO}_3$  followed with brine. Purification on a short silica gel column (30% to 50% EtOAc in hexanes) yielded 24.5 g (95%) of the title compound.

5

Step 16: Methyl 1-(((3-(2-(2-hydroxy-2-propyl)-phenyl)-1(R)-(3-((2-tetrahydropyranyl)-oxymethyl)phenyl)propyl)thio)methyl)-cyclopropaneacetate

10

The diol of Step 15 (17.9 g, 46.6 mmol) was dissolved in  $\text{CH}_3\text{CN}$  (40 mL) and DMF (10 mL) and cooled to  $-42^\circ\text{C}$  under nitrogen. Diisopropylethylamine (8.5 mL, 48.9 mmol) was added followed by methanesulphonyl chloride (3.6 mL, 46.6 mmol) dropwise. The solution was stirred 1.5 hours with a mechanical stirrer while maintaining the temperature between  $-42^\circ\text{C}$  and  $-35^\circ\text{C}$ ; then it was cooled to  $-45^\circ\text{C}$ . The thiol of Step 10 (7.84 g, 48.9 mmol) was added followed by dropwise addition of DMF (15 mL). The potassium tert-butoxide in THF solution (56 mL, 1.75M, 97.9 mmol) was added to the reaction mixture within 20 minutes using a syringe pump. Stirring continued for 5 hours with slow warming from  $-35^\circ\text{C}$  to  $-22^\circ\text{C}$ , giving a very thick translucent gel. The reaction was quenched with saturated aq.  $\text{NH}_4\text{Cl}$  (250 mL) and EtOAc (300 mL). The product was extracted with EtOAc, washed with  $\text{H}_2\text{O}$  and brine, and dried over  $\text{MgSO}_4$ . Purification by flash chromatography (20% to 30% EtOAc in hexanes) gave 16.8 g (68%) of the title compound.

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Step 17: (R)Methyl 1-(((1-(3-(hydroxymethyl)-phenyl)-3-(2-(2-hydroxy-2-propyl)phenyl)-propyl)thio)methyl)cyclopropaneacetate

5 To the hydroxy ester from Step 16 (9.02 g, 17.1 mmol) in anhydrous methanol (60 mL) under nitrogen was added pyridine (50  $\mu$ L) followed by pyridinium p-toluenesulfonate (1.1 g, 4.3 mmol). The reaction mixture was stirred 3.5 hours at 55°C, then  
10 at r.t. overnight before concentrating in vacuo. The residue was diluted with EtOAc (500 mL) and washed with H<sub>2</sub>O, saturated aq. NaHCO<sub>3</sub>, NaH<sub>2</sub>PO<sub>4</sub> buffer (pH = 4.5) and with brine. After drying over MgSO<sub>4</sub> and evaporation of the solvents, the product was  
15 purified by flash chromatography (40% to 60% EtOAc in hexanes) giving 6.85 g (91%) of the title compound. <sup>1</sup>H NMR (CD<sub>3</sub>COCD<sub>3</sub>):  $\delta$  7.41 (2H, m), 7.27 (3H, m), 7.09 (3H, m), 4.63 (2H, d), 4.19 (1H, t), 3.95 (1H, t), 3.88 (1H, s), 3.57 (3H, s), 3.1 (1H, ddd), 2.8 (1H, ddd), 2.5 (2H, s), 2.4 (2H, d), 2.17 (2H, m), 1.52  
20 (6H, s), 0.52-0.35 (4H, m).

Step 18: (R) Methyl 1-(((1-(3-formylphenyl)-3-(2-(2-hydroxy-2-propyl)phenyl)propyl)thio)methyl)  
25 cyclopropaneacetate

To the dihydroxy-ester from Step 17 (6.8 g, 15.4 mmol) in EtOAc (150 mL) at 50°C was added manganese dioxide (6.7 g, 76.8 mmol). After stirring  
30 for 30 minutes at 50°C more MnO<sub>2</sub> (6.7 g) was added, and 30 minutes later, a third portion of MnO<sub>2</sub> (6.7 g) was added. An hour later, the warm reaction mixture

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was filtered through celite and the cake was washed with additional EtOAc. Evaporation of the solvents gave 5.62 g (83%) of the desired aldehyde.

<sup>1</sup>H NMR (CD<sub>3</sub>COCD<sub>3</sub>): δ 10.4 (1H, s), 7.9 (1H, brs), 7.8 (2H, m), 7.58 (1H, t), 7.38 (1H, brd), 7.1 (3H, m), 4.1 (1H, t), 3.54 (3H, s), 3.13 (1H, ddd), 2.85 (1H, ddd), 2.51 (2H, s), 2.49 (2H, d), 2.2 (2H, m), 1.51 (6H, s), 0.52-0.32 (4H, m).

10    Step 19:    (R)-Methyl 1-(((1-(3-(2-(5,6-dimethyl-2-pyridinyl)ethenyl)phenyl)-3-(2-(2-hydroxy-2-propyl)phenyl)propyl)thio)methyl)cyclopropanacetate

15            To a suspension of the phosphonium salt from Step 6 (924 mg, 2.0 mmol) in dry THF (10 mL) at -78°C was added n-BuLi (800 µL, 2.5M in hexanes, 2 mmol). The orange mixture was stirred 30 min. at -78°C, warmed to ~0°C for 15 min. and then cooled to -78°C.

20    The aldehyde from Step 18 (2 mL, 0.5M/THF, 1 mmol) was added and the resulting mixture was allowed to stir for 2 hours with slow warming to r.t. Saturated aq. NH<sub>4</sub>Cl was added and the mixture was extracted with EtOAc (2x). The organic extracts were washed

25    with brine, dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated to an oil. Purification of the crude oil by flash chromatography (30% EtOAc in hexanes) gave 423 mg (78%) of the title compound.

<sup>1</sup>H NMR (CD<sub>3</sub>COCD<sub>3</sub>): δ 7.65-7.0 (12H, m), 4.0 (1H, t), 3.85 (1H, s), 3.53 (3H, s), 3.12 (1H, ddd), 2.85 (1H, ddd), 2.51 (2H, s), 2.45 (3H, s), 2.40 (2H, dd), 2.26 (3H, s), 2.20 (2H, m), 1.50 (6H, s), 0.52-0.35 (4H, m).

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Step 20: (R)-Sodium 1-(((1-(3-(2-(5,6-dimethyl-2-pyridinyl)ethenyl)phenyl)-3-(2-(2-hydroxy-2-propyl)phenyl)propyl)thio)methyl)cyclopropaneacetate

5

To a solution of ester from Step 19 (423 mg, 0.78 mmol) in MeOH (50 mL) and THF (10 mL) was added aqueous 2N NaOH (800  $\mu$ L, 1.6 mmol). The mixture was stirred overnight at r.t. Saturated aq.  $\text{NH}_4\text{Cl}$  was added and the mixture extracted with EtOAc (3x). The combined organic layers were washed with brine, dried over  $\text{Na}_2\text{SO}_4$  and concentrated to an oil. Purification of this crude by flash chromatography (50% EtOAc in hexanes, then 1% AcOH in 70% EtOAc in hexanes) gave 15 382 mg (89%) of the corresponding acid. To this acid in EtOH (4 mL) and THF (1 mL) was added 2N NaOH (350  $\mu$ L, 0.7 mmol). The solvents were evaporated and the product was lyophilysed to give the title compound.

20 acid  $^1\text{H}$  NMR ( $\text{CD}_3\text{COCD}_3$ ):  $\delta$  7.7-7.0 (12H, m), 4.05 (1H, t), 3.12 (1H, ddd), 2.87 (1H, ddd), 2.57 (2H, s), 2.48 (3H, s), 2.43 (1H, brs), 2.29 (3H, s), 2.20 (2H, m), 1.52 (6H, s), 0.55-0.35 (4H, m).

sodium salt  $^1\text{H}$  NMR ( $\text{CD}_3\text{SOCD}_3$ ):  $\delta$  8.1-7.45 (12H, m), 25 4.97 (1H, s), 4.40 (1H, t), 3.5 (1H, ddd), 3.15 (1H, ddd), 3.1-2.9 (2H, m), 2.9 (3H, s), 2.7 (3H, s), 2.7-2.4 (4H, m), 1.90 (3H, s), 1.88 (3H, s), 0.82 (2H, m), 0.65 (2H, m).

Analysis calculated for  $\text{C}_{33}\text{H}_{38}\text{NO}_3\text{SNa}\cdot\text{H}_2\text{O}$ :

30 C, 69.57; H, 7.08; N, 2.46

Found: C, 69.10; H, 7.16; N, 2.47

Mass spec. (FAB):  $[\text{M} + 23]^+$  at 574.4 (100%),  $\text{MH}^+$  at 552.3 (21%)

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EXAMPLE 2

(R)-Sodium 1-(((1-(3-(2-(5-trifluoromethyl-2-pyridinyl)ethenyl)phenyl)-3-(2-(2-hydroxy-2-propyl)-phenyl)propyl)thio)methyl)cyclopropaneacetate

Step 1: 2-Methyl-5-(trifluoromethyl)pyridine

To a slurry of polymer supported tetrakis triphenylphosphine palladium (4.0 g, catalytic) in dioxane (40 mL) at r.t. was added 2-chloro-5-trifluoromethylpyridine (14.7 g, 81 mmol) followed by trimethylaluminum (32.5 mL, 2M in hexanes, 65 mmol). The mixture was stirred 6 hours at r.t., then heated to reflux overnight. After cooling to ~25°C, the mixture was slowly poured onto (500 g) ice, stirred 10 min. and then tartaric acid (5 g) and sodium potassium tartrate (25 g) were added. Stirring was continued for 25 min. before filtering off solids. The filtrate was extracted with CH<sub>2</sub>Cl<sub>2</sub> (4x 200 mL) and the solvents were evaporated. Distillation of the residue under vacuum (20 mm Hg) afforded 4.2 g of a liquid. <sup>1</sup>H NMR showed it to be a 1:5 mixture of dioxane and the title compound. This mixture was used without further purification.

Step 2: 2-Methyl-5(trifluoromethyl)pyridine N-oxide

To a solution of crude 2-methyl-5-trifluoromethylpyridine from Step 1 (3.8 g) in CHCl<sub>3</sub> (40 mL) was added m-chloroperbenzoic acid (3.0 g, 17.4 mmol) and the resulting mixture was stirred 5 hours at

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r.t. More peracid (1.5 g, 8.7 mmol) was added and the reaction stirred another 2 hours. Calcium hydroxide (3.0 g, 40.5 mmol) was added and the slurry was vigorously stirred 20 minutes before filtering  
5 through celite. Evaporation of the solvents gave 4.5 g of the title compound containing a little dioxane and  $\text{CHCl}_3$ . This material was used without further purification.

$^1\text{H}$  NMR ( $\text{CD}_3\text{COCD}_3$ ):  $\delta$  8.53 (1H, brs), 7.68 (1H, d),  
10 7.52 (1H, d), 2.42 (3H, s).

Step 3: ((5-Trifluoromethyl)-2-pyridinyl)methyl)-  
triphenylphosphonium chloride

15 To a solution of the pyridine N-oxide from Step 2 (4.5 g crude) in  $\text{CH}_2\text{Cl}_2$  (30 mL) was added phosphoryl chloride (230  $\mu\text{L}$ , 2.5 mmol) at r.t. Then triethylamine (3.5 mL, 25 mmol) and phosphoryl  
20 chloride (2.10 mL, 22.5 mmol) were added simultaneously at such a rate in order to cause the solution to reflux. After the addition, the mixture was heated to reflux for 1.5 hours, then stirred at  
25 r.t. overnight. Saturated aq.  $\text{NaHCO}_3$  (75 mL), 25% aq.  $\text{NH}_4\text{OAc}$  (50 mL) and  $\text{CH}_2\text{Cl}_2$  (75 mL) were added and the resulting mixture was vigorously stirred for 15  
30 minutes. The crude was obtained after extraction with  $\text{CH}_2\text{Cl}_2$  (2x), drying over  $\text{MgSO}_4$  and evaporation of the solvents. Purification by flash chromatography ( $\text{CH}_2\text{Cl}_2$ ) gave after partial  
evaporation of the solvent a  $\text{CH}_2\text{Cl}_2$  solution of the desired 2-chloromethyl-5-trifluoromethylpyridine. To this was added triphenylphosphine (6.5 g, 25 mmol)

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and acetonitrile (10 mL). The mixture was heated to reflux for 2 hours ( $\text{CH}_2\text{Cl}_2$  distills off), stirred at r.t. overnight and concentrated by distillation.

Evaporation to dryness left a solid that was swished in  $\text{Et}_2\text{O}$  (2x) affording 3.0 g of the title compound.

$^1\text{H}$  NMR ( $\text{CDCl}_3$ ):  $\delta$  8.5 (1H, brs), 8.32 (1H, d), 7.92-7.55 (16H, m), 6.02 (2H, d).

Step 4: (R)-Sodium 1-(((1-(3-(2-(5-trifluoromethyl-2-pyridinyl)ethenyl)phenyl)-3-(2-(2-hydroxy-2-propyl)phenyl)propyl)thio)methyl)cyclopropaneacetate

Using the procedure described in Steps 7-20 of Example 1, the phosphonium salt of Step 3 was converted to the title compound.

Analysis calculated for  $\text{C}_{32}\text{H}_{33}\text{NO}_3\text{SF}_3\text{Na}\cdot\text{H}_2\text{O}$ :

C, 63.04; H, 5.79; N, 2.30

Found: C, 60.50; H, 5.76; N, 2.11.

Mass spec. (FAB):  $[\text{M}+23]^+$  at 614 (100%),  $\text{MH}^+$  at 592 (57%)

$^1\text{H}$  NMR ( $\text{CD}_3\text{COCD}_3$ ):  $\delta$  8.88 (1H, s), 8.07 (1H, brd), 7.88 (1H, d), 7.73 (2H, m), 7.55-7.3 (5H, m), 7.05 (3H, m), 4.07 (1H, t), 3.3-2.6 (5H, m), 2.4-2.05 (4H, m), 1.56 (3H, s), 1.51 (3H, s), 0.45 (2H, m), 0.25 (2H, m).

#### EXAMPLES 3-5

Using the method described in Example 2 (Steps 2-4), starting from 6-chloro-2-picoline, 5-n-butyl-2-picoline and from 2,4,6-collidine, the compounds of Examples 3-5 were prepared.

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Example 3: Exact mass found for  $C_{31}H_{33}SO_3NCINa$   
(M+1): 558.18465

Calculated: 558.1845635

5 Example 4: Mass spec. (FAB):  $[M+23]^+$  at 602 (100%),  
 $MH^+$  at 580 (32%)

Example 5: Mass spec. (FAB):  $[M+23]^+$  at 574 (57%),  
 $MH^+$  at 552 (20%)

#### EXAMPLE 6

10

Starting from 2-picoly1 chloride and using  
the same method described in Example 1 (Steps 6-20),  
the compound of Example 6 was prepared.

15 Mass spec. (FAB):  $[M+23]^+$  at 546 (33%),  $MH^+$  at 524  
(44%).

#### EXAMPLE 7

20 (R)-Sodium 1-(((1-(3-(2-(5-methoxy-2-pyridinyl)-  
ethenyl)phenyl)-3-(2-(2-hydroxy-2-propyl)phenyl)-  
propyl)thio)methyl)cyclopropaneacetate

#### Step 1: 5-Methoxy-2-picoline

25 Sodium hydride (3.6 g, 150 mmol) was added  
portionwise to 5-hydroxy-2-picoline (15.0 g, 137  
mmol) in DMF (120 mL). After 30 min.,  $CH_3I$  (10.2 mL,  
156 mmol) was added and the mixture stirred 2 hours  
at r.t. Water was added and the compound was  
30 extracted with  $Et_2O$  (3x). The organic extracts were  
washed with brine (2x), dried over  $Na_2SO_4$  and  
concentrated in vacuo to give 6.7 g of the title  
compound.

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Step 2: 5-Methoxy-2-picoline N-oxide

5 The 5-methoxy-2-picoline from Step 1 (6.7 g) was treated with 30% H<sub>2</sub>O<sub>2</sub> (6.12 mL) in acetic acid (40 mL) at 100°C overnight. After cooling to r.t., excess MnO<sub>2</sub> was added and the slurry was stirred for 2 hours. Filtration and concentration in vacuo gave 7.0 g of the title N-oxide.

10 Step 3: (R)-Sodium 1-(((1-(3-(2-(5-methoxy-2-pyridinyl)ethenyl)phenyl)-3-(2-(2-hydroxy-2-propyl)phenyl)propyl)thio)methyl)cyclopropaneacetate

15 Using the same procedure described in Steps 3-4 of Example 2, the pyridine N-oxide of Step 2 was converted to the title compound. Exact mass found for C<sub>32</sub>H<sub>36</sub>NO<sub>4</sub>SNa(M+1): 554, 23414  
Calculated: 554, 23410

20

EXAMPLE 8

(R)-Sodium 1-(((1-(3-(2-(5,6-cyclohexeno-2-pyridinyl)-ethenyl)phenyl)-3-(2-(2-hydroxy-2-propyl)phenyl)-  
25 propyl)thio)methyl)cyclopropaneacetate

Step 1: 2-(Acetyoxymethyl)quinoline

30 Sodium acetate (38.3 g, 467 mmol), Cs<sub>2</sub>CO<sub>3</sub> (38.0 g, 116.7 mmol) and 2-chloromethylquinolinium hydrochloride (25.0 g, 116.7 mol) (Eur. Pat. Appl. 284174, 28 Sept. 1988) were mixed together in DMF

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(200 mL) and stirred overnight at 65°C. The reaction was quenched with saturated aq.  $\text{NH}_4\text{Cl}$ , the crude compound was extracted with EtOAc (3x), washed with brine and dried over  $\text{MgSO}_4$ . Concentration in vacuo and purification of the residue by flash chromatography (from 10% to 20% EtOAc in toluene) gave 22.11 g (94%) of the title compound.

$^1\text{H}$  NMR ( $\text{CD}_3\text{COCD}_3$ ):  $\delta$  8.32 (1H, d), 8.01 (1H, d), 7.95 (1H, d), 7.75 (1H, dt), 7.58 (2H, m), 5.33 (2H, s), 2.15 (3H, s).

Step 2: 6-Hydroxymethyl-2,3-cyclohexenopyridine

The 2-acetoxymethylquinoline from Step 1 (22.0 g, 109 mmol) was dissolved in TFA (100 mL) in a Parr pressure bottle.  $\text{PtO}_2$  (1 g, 4.4 mmol) was added and the mixture was hydrogenated at 50 psi of hydrogen for 110 minutes. After concentrating overnight with a nitrogen stream and filtering off the catalyst, the residue was diluted with MeOH (100 mL) before addition of excess 10N NaOH (~20 mL). This solution was stirred 15 minutes, then quenched with solid  $\text{NH}_4\text{Cl}$  and saturated aq.  $\text{NH}_4\text{Cl}$  and extracted with EtOAc (3x). The organic layers were washed with brine, dried over  $\text{MgSO}_4$  and concentrated to an oily solid. Purification by flash chromatography (from 30% to 50% acetone in  $\text{CH}_2\text{Cl}_2$ ) gave a solid that was swished in  $\text{Et}_2\text{O}$  yielding 7.5 g (42%) of the title compound as an oily solid.

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$^1\text{H}$  NMR ( $\text{CD}_3\text{COCD}_3$ ):  $\delta$  7.40 (1H, d), 7.15 (1H, d), 4.59 (2H, s), 4.40 (1H, brs), 2.80 (2H, t), 2.75 (2H, t), 1.9-1.7 (4H, m).

5     Step 3:     ((5,6-Cyclohexeno-2-pyridinyl)methyl)tri-  
                  phenylphosphonium methanesulfonate

                  Methanesulfonyl chloride (3.26 mL, 42.1 mmol) was added to a solution of hydroxymethyl-  
10     pyridine from Step 2 (5.5 g, 33.7 mmol) and triethylamine (6.1 mL, 43.8 mmol) in  $\text{CH}_2\text{Cl}_2$  at  $-40^\circ\text{C}$ . After stirring 2 hours with slow warming to  $-20^\circ\text{C}$ , saturated aq.  $\text{NaHCO}_3$  was added and the mixture was stirred 15 minutes before separation of the  
15     layers. The aqueous phase was extracted with EtOAc (2x) and the combined organic layers were washed with brine, dried over  $\text{MgSO}_4$  and concentrated to give the corresponding crude mesylate (8.5 g). This oil was dissolved in acetonitrile (150 mL), triphenyl-  
20     phosphine (14.1 g, 53.9 mmol) was added and the solution was refluxed for 2 hours. Upon cooling and evaporation of the solvent, the solid residue was swished (2x) in  $\text{Et}_2\text{O}$  to give 16.5 g (97%) of the title compound.

25      $^1\text{H}$  NMR ( $\text{CDCl}_2$ ):  $\delta$  7.85-7.55 (15H, m), 7.50 (1H, d), 7.23 (1H, d), 5.32 (2H, d), 2.70 (3H, s), 2.62 (2H, m), 2.39 (2H, brt), 1.70 (4H, m).

30     Step 4:     (R)-Sodium 1-(((1-(3-(2-(5,6-cyclohexeno-2-pyridinyl)ethenyl)phenyl)-3-(2-(2-hydroxy-2-propyl)phenyl)propyl)thio)methyl)  
                  cyclopropaneacetate

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Using the procedure described in Steps 7-20 of Example 1, the title compound was prepared from the phosphonium salt of Step 3.

Analysis calculated for  $C_{35}H_{40}NO_3SNa \cdot 1\frac{1}{2}H_2O$ :

5 C, 69.51; H, 7.17; N, 2.32

Found: C, 69.17; H, 7.15; N, 2.11

Mass spec. (FAB):  $[M+23]^+$  at 600 (26%),  $MH^+$  at 578 (33%)

10  $^1H$  NMR ( $CD_3COCD_3$ ):  $\delta$  7.75-7.0 (12H, m), 4.02 (1H, brt), 3.25-2.6 (9H, m), 2.22 (2H, brs), 2.1 (2H, m), 1.8 (4H, m), 1.56 (3H, s), 1.51 (3H, s), 0.43 (2H, m), 0.25 (2H, m)

#### EXAMPLE 9

15

(R)-Sodium 1-(((1-(3-(2-(5,6-cyclopenteno-2-pyridinyl)ethenyl)phenyl)-3-(2-hydroxy-2-propyl)phenyl)propyl)-thio)methyl)cyclopropaneacetate

20 Step 1: 6-Methyl-2,3-cyclopentenopyridine

Wet Raney Nickel (15 g) was heated at 135°C in a mixture of dodecane (130 mL) and 1-octanol (70 mL, 440 mmol) for 30 minutes to remove most of the H<sub>2</sub>O (Dean Stark trap). 2,3-cyclopentenopyridine (15.0 g, 126 mmol) was added and the resulting mixture was heated at 185°C overnight. More Ra/Ni (5 g) and 1-octanol (15 mL, 94 mmol) were added and heating at 190°C was continued for another 24 hours. 25 After cooling to r.t., H<sub>2</sub>O (150 mL) and hexanes (300 mL) were added; the aqueous phase was extracted with hexanes (3x), the combined organic layers were

30

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washed with 6N HCl (50 mL) followed by 1N HCl (40 mL). To this acid phase was added 10N NaOH until basic. Extraction with CH<sub>2</sub>Cl<sub>2</sub> (3x), drying over MgSO<sub>4</sub> and evaporation of the solvents gave 15.2 g of the title compound containing ~10% of starting material.

<sup>1</sup>H NMR (CDCl<sub>3</sub>): δ 7.49 (1H, d), 6.90 (1H, d), 3.0 (2H, t), 2.9 (2H, t), 2.52 (3H, s), 2.12 (2H, m).

10 Step 2: (R)-Sodium 1-(((1-(3-(2-(5,6-cyclopenteno-2-pyridinyl)ethenyl)phenyl)-3-(2-hydroxy-2-propyl)phenyl)propyl)thio)methyl)cyclopropaneacetate

15 Using the procedure described in Steps 2-4 of Example 2, the pyridine of Step 1 was converted into the title compound.

Analysis calculated for C<sub>34</sub>H<sub>38</sub>NO<sub>3</sub>SNa•2H<sub>2</sub>O:

20 C, 68.09; H, 7.06; N, 2.34

Found: C, 65.71; H, 6.72; N, 2.01

Mass spec. (FAB): [M+23]<sup>+</sup> at 586 (20%), MH<sup>+</sup> at 564 (31%)

25 <sup>1</sup>H NMR (CD<sub>3</sub>COCD<sub>3</sub>): δ 7.7-7.0 (12H, m), 4.20 (1H, t), 3.2 (1H, ddd), 2.95-2.7 (9H, m), 2.6 (2H, dd), 2.3-2.0 (3H, m), 1.56 (3H, s), 1.51 (3H, s), 0.42 (2H, m), 0.23 (2H, m).

#### EXAMPLE 10

30

Starting from 3-phenylpyridine and using the same procedure as described in Example 9, the compound of Example 10 was prepared.

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Mass spec. (FAB):  $[M+23]^+$  at 622 (29%),  $MH^+$  at 600 (13%).

EXAMPLE 11

5

(R)-Sodium 1-(((1-(3-(2-(6-isopropyl-2-pyridinyl)-ethenyl)phenyl)-3-(2-(2-hydroxy-2-propyl)phenyl)-propyl)thio)methyl)cyclopropaneacetate

10    Step 1:    6-Ethyl-2-pyridinecarboxaldehyde

To a solution of N,N,N-trimethylethylenediamine (4.29 mL, 33 mmol) in THF (20 mL) at -25°C was added n-BuLi (13.2 mL, 2.5M in hexanes, 33 mmol). This solution was stirred 15 minutes, then transferred into a -78°C solution of 6-methyl-2-pyridine-carboxaldehyde (3.63 g, 30 mmol) in THF (80 mL). After stirring 30 minutes at -78°C, a THF (50 mL) solution of lithium diisopropylamide (33 mmol) was added and the now dark red mixture was stirred 1 hour at -78°C. Methyl iodide (5.68 g, 40 mmol) in THF (10 mL) was added and reaction mixture was allowed to warm to r.t. for 4 hours. Water and 25% aq.  $NH_4OAc$  were added, the crude product was extracted with EtOAc (2x) and solvents were evaporated. Purification by flash chromatography (from 5% to 10% EtOAc in hexanes) gave 660 mg of the title compound.

30    Step 2:    6-Isopropyl-2-pyridinecarboxaldehyde

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The alkylation of 6-ethyl-2-pyridine-carboxaldehyde from Step 1 was performed as described above for 6-methyl,2-pyridinecarboxaldehyde.

5  $^1\text{H}$  NMR ( $\text{CD}_3\text{COCD}_3$ ):  $\delta$  10.0 (1H, s), 7.93 (1H, t), 7.75 (1H, d), 7.57 (1H, d), 3.18 (1H, m), 1.32 (6H, d).

Step 3: 6-Isopropyl-2-(hydroxymethyl)pyridine

10  $\text{NaBH}_4$  (37 mg, 1 mmol) was added to a MeOH (400  $\mu\text{L}$ ) and THF (4 mL) solution of 6-isopropyl-2-pyridinecarboxaldehyde from Step 2 (160 mg, 1.07 mmol) at r.t. and the resulting mixture was stirred for 1 hour. Saturated aq.  $\text{NH}_4\text{Cl}$  was added and extraction with EtOAc (2x) gave, after evaporation of  
15 the solvents, 156 mg (97%) of the title compound which was used without further purification.

Step 4: (R)-Sodium 1-(((1-(3-(2-(6-isopropyl-2-pyridinyl)ethenyl)phenyl)-3-(2-(2-hydroxy-2-propyl)phenyl)propyl)thio)methyl)cyclopropaneacetate

20

Using the method described in Steps 5-20 of Example 1, the hydroxymethylpyridine of Step 3 was  
25 converted to the title compound.

Mass spec. (FaB):  $[\text{M}+23]^+$  at 588 (100%),  $\text{MH}^+$  at 566 (33%)

acid  $^1\text{H}$  NMR ( $\text{CD}_3\text{COCD}_3$ ):  $\delta$  7.8-7.05 (13H, m), 4.05 (1H, t), 3.17 (1H, ddd), 3.07 (2H, m), 2.88 (1H, ddd), 2.6 (2H, s), 2.43 (2H, s), 2.2 (2H, m), 1.55 (6H, s), 1.30 (6H, d), 0.55-0.35 (4H, m).  
30

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EXAMPLE 12

(R)-Sodium 1-(((1-(3-(2-(5-ethyl-6-methyl-2-pyridinyl)ethenyl)phenyl)-3-(2-(2-hydroxy-2-propyl)phenyl)-propyl)thio)methyl)cyclopropaneacetate

Step 1: 5-Ethyl-6-methyl-2-pyridinecarbonitrile

Potassium bis(trimethylsilyl)amide (41 mL, 0.5M in toluene, 20.4 mmol) was added to a solution of 2-cyano-5,6-dimethylpyridine from Example 1, Step 2 (2.45 g, 18.5 mmol) and hexamethylphosphoric triamide (9.7 mL, 55.6 mmol) in dry THF (30 mL) at -78°C. After stirring 30 minutes, MeI (5.77 mL, 92.7 mmol) was added and the reaction mixture stirred 45 minutes at -78°C. Saturated aq. NH<sub>4</sub>Cl was added, the mixture was extracted with EtOAc (3x), the organic layers were dried over MgSO<sub>4</sub> and concentrated in vacuo. The residue was purified by flash chromatography (from 15:15:70 to 20:20:60 of Et<sub>2</sub>O: CH<sub>2</sub>Cl<sub>2</sub>: hexanes) to give 1.2 g (44%) of the title compound.

Step 2: Methyl 5-ethyl-6-methyl-2-pyridine-carboxylate

Dry HCl (gas) was bubbled through a MeOH (30 mL) solution of 2-cyanopyridine from Step 1 (1.4 g, 9.6 mmol) at 0°C until saturation. The flask was sealed with a new rubber septum and the reaction mixture was stirred 30 hours at r.t. After careful depressurisation, H<sub>2</sub>O (5 mL) was added and the MeOH was evaporated. The aqueous residue was

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neutralized with saturated aq. NaHCO<sub>3</sub> and extracted with EtOAc (3x). The organic extracts were dried over MgSO<sub>4</sub>, concentrated in vacuo, and the residue was purified by flash chromatography (from 30% to 50% EtOAc in hexanes) to give 1.49 g (87%) of the title compound.

<sup>1</sup>H NMR (CD<sub>3</sub>COCD<sub>3</sub>): δ 7.85 (1H, d), 7.67 (1H, d), 3.86 (3H, s), 2.72 (2H, q), 2.58 (3H, s), 1.23 (3H, t).

10    Step 3:    (R)-Sodium 1-(((1-(3-(2-(5-ethyl-6-methyl-2-pyridinyl)ethenyl)phenyl)-3-(2-(2-hydroxy-2-propyl)phenyl)propyl)thio)-methyl)cyclopropaneacetate

15            Using the procedure described in Steps 4-20 of Example 1, the methyl ester of Step 2 was converted to the title compound.

acid <sup>1</sup>H NMR (CD<sub>3</sub>COCD<sub>3</sub>): δ 7.7-7.0 (12H, m), 4.05 (1H, t), 3.15 (1H, ddd), 2.75 (2H, m), 2.66 (2H, q), 2.58 (2H, s), 2.51 (3H, s), 2.43 (2H, d), 2.21 (2H, m), 1.52 (6H, s), 1.21 (3H, t), 0.55-0.4 (4H, m).

### EXAMPLE 13

25    (R)-Sodium 1-(((1-(3-(2-(6-butyl-2-pyridinyl)-ethenyl)phenyl)-3-(2-(2-hydroxy-2-propyl)phenyl)-propyl)thio)methyl)cyclopropaneacetate

Step 1:    6-(1-Butenyl)-2-picoline

30

n-Butyllithium (7.5 mL, 1.6M/hexanes, 12 mmol) was added to a slurry of propyltriphenylphosphonium bromide (4.62 g, 12 mmol) in THF (40 mL)

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at -78°C. The mixture was warmed to 0°C and stirred 30 min., then cooled again to -78°C. A THF (10 mL) solution of 6-methyl-2-pyridinecarboxaldehyde (1.21 g, 10 mmol) was added dropwise. The reaction mixture  
5 was allowed to warm to r.t. and stirred for 2 hours. Silica gel was added and the slurry was filtered on a short (SiO<sub>2</sub>) column eluting with 10% EtOAc in hexanes to give 990 mg of the title compound.

10 Step 2: 6-Butyl-2-picoline

The picoline from Step 1 (970 mg) was hydrogenated for 50 minutes at 20 psi of hydrogen in EtOAc (30 mL) using 5% Pd/C (90 mg) as catalyst.  
15 Filtration and concentration in vacuo gave 910 mg of the title compound.

20 Step 3: (R)-Sodium 1-(((1-(3-(2-(6-butyl-2-pyridinyl)ethenyl)phenyl)-3-(2-(2-hydroxy-2-propyl)phenyl)propyl)thio)methyl)cyclopropaneacetate

Using the procedure described in Steps 2-4 of Example 2, the picoline of Step 2 was converted to  
25 the title compound.

Elemental Analysis calculated for C<sub>35</sub>H<sub>42</sub>NO<sub>3</sub>SNa•H<sub>2</sub>O:

C, 70.32; H, 7.42; N, 2.34

Found: C, 70.34; H, 7.35; N, 1.86

Mass spec. (FAB): [M + 23]<sup>+</sup> at 602 (80%), MH<sup>+</sup> at 580  
30 (78%)

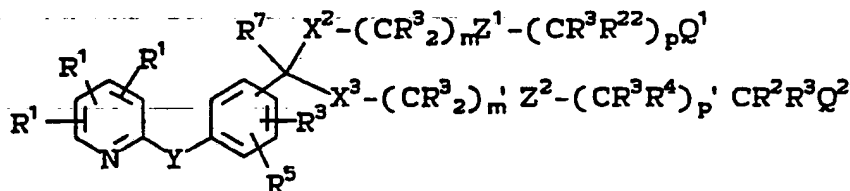
acid <sup>1</sup>H NMR (CD<sub>3</sub>COCD<sub>3</sub>): δ 7.7-7.0 (13H, m), 4.05 (1H, t), 3.13 (1H, ddd), 2.88 (2H, m), 2.74 (2H, t), 2.56 (2H, s), 2.42 (2H, brs), 2.2 (2H, m), 1.72 (2H, m), 1.52 (6H, s), 1.40 (2H, m), 0.92 (3H, t), 0.55-0.35 (4H, m).

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WHAT IS CLAIMED IS:

1. A compound of the Formula:



wherein:

15  $R^1$  is H, halogen, CN, lower alkyl, cyloalkyl, polyhalo lower alkyl, lower alkoxy, lower alkoxy lower alkyl, lower alkylthio lower alkyl, lower alkenyl, substituted or unsubstituted phenyl, pyridyl, thiazolyl, oxazolyl, furanyl or thienyl, or adjacent  $R^1$ 's and the carbons through which they are attached may form a saturated ring of 5 to 20 carbon atoms;

25  $R^2$  is lower alkyl, lower alkenyl, lower alkynyl,  $-CF_3$ ,  $-CH_2F$ ,  $-CHF_2$ ,  $-CH_2CF_3$ , substituted or unsubstituted phenyl, substituted or unsubstituted benzyl, substituted or unsubstituted 2-phenethyl, or two  $R^2$  groups joined to the same carbon may form a saturated ring of up to 8 members containing 0 to 2 heteroatoms chosen from O, S, and N;

30  $R^3$  is H or  $R^2$ ;  
 $CR^3R^{22}$  may be the radical of a standard amino acid;  
 $R^4$  is halogen,  $-NO_2$ ,  $-CN$ ,  $-OR^3$ ,  $-SR^3$ ,  $NR^3R^3$ ,  $NR^3C(O)R^7$ , or  $R^3$ ;

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- $R^5$  is H, halogen,  $-\text{NO}_2$ ,  $-\text{N}_3$ ,  $-\text{CN}$ ,  $-\text{SR}^2$ ,  $-\text{NR}^3\text{R}^3$ ,  
 $-\text{OR}^3$ , lower alkyl, or  $-\text{C}(\text{O})\text{R}^3$ ;
- $R^6$  is  $-(\text{CH}_2)_s-\text{C}(\text{R}^7\text{R}^7)-(\text{CH}_2)_s-\text{R}^8$  or  
 $-\text{CH}_2\text{C}(\text{O})\text{NR}^{12}\text{R}^{12}$ ;
- 5  $R^7$  is H or lower alkyl;
- $R^8$  is A) a monocyclic or bicyclic heterocyclic  
radical containing from 3 to 12 nuclear  
carbon atoms and 1 or 2 nuclear heteroatoms  
selected from N, S or O and with each ring  
10 in the heterocyclic radical being formed of  
5 or 6 atoms, or  
B) the radical  $\text{W}-\text{R}^9$ ;
- $R^9$  contains up to 20 carbon atoms and is (1) an  
alkyl group or (2) an alkylcarbonyl group of  
15 an organic acyclic or monocyclic carboxylic  
acid containing not more than 1 heteroatom  
in the ring;
- $R^{10}$  is  $-\text{SR}^{11}$ ,  $-\text{OR}^{12}$ , or  $-\text{NR}^{12}\text{R}^{12}$ ;
- $R^{11}$  is lower alkyl,  $-\text{C}(\text{O})\text{R}^{14}$ , unsubstituted phenyl,  
20 or unsubstituted benzyl;
- $R^{12}$  is H,  $\text{R}^{11}$ , or two  $\text{R}^{12}$  groups joined to the same  
N may form a saturated ring of 5 or 6  
members containing up to two heteroatoms  
chosen from O, S, and N;
- 25  $R^{13}$  is lower alkyl, lower alkenyl, lower alkynyl,  
 $-\text{CF}_3$ , or substituted or unsubstituted  
phenyl, benzyl, or 2-phenethyl;
- $R^{14}$  is H or  $\text{R}^{13}$ ;
- $R^{15}$  is  $\text{R}^3$  or halogen;
- 30  $R^{16}$  is H, lower alkyl, or OH;
- $R^{17}$  is lower alkyl, lower alkenyl, lower alkynyl,  
or substituted or unsubstituted phenyl,  
benzyl, or 2-phenethyl;

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- $R^{18}$  is lower alkyl, lower alkenyl, lower alkynyl,  
 $-CF^3$ , or substituted or unsubstituted  
phenyl, benzyl, or 2-phenethyl;  
 $R^{19}$  is lower alkyl, lower alkenyl, lower alkynyl,  
5  $-CF^3$ , or substituted or unsubstituted  
phenyl, benzyl, or 2-phenethyl;  
 $R^{20}$  is H, lower alkyl, substituted or unsubstituted  
phenyl, benzyl, phenethyl, or pyridinyl, or  
10 two  $R^{20}$  groups joined to the same N may form  
a saturated ring of 5 or 6 members  
containing one to two heteroatoms chosen  
from O, S, and N;  
 $R^{21}$  is H or  $R^{17}$ ;  
 $R^{22}$  is  $R^4$ ,  $CHR^7OR^3$ , or  $CHR^7SR^2$ ;  
15 m and m' are independently 0-8;  
p and p' are independently 0-8;  
m + p is 1-10 when  $X^2$  is O, S,  $S(O)$ , or  $S(O)_2$ ;  
m + p is 0-10 when  $X^2$  is  $CR^3R^{16}$  or a bond;  
m' + p' is 0-10;  
20 s is 0-3;  
 $Q^1$  is  $-C(O)OR^3$ , 1H (or 2H)-tetrazol-5-yl,  
 $-C(O)OR^6$ ,  $-C(O)NHS(O)_2R^{13}$ ,  $-CN$ ,  
 $-C(O)NR^{12}R^{12}$ ,  $NR^{21}S(O)_2R^{13}$ ,  
 $-NR^{12}C(O)NR^{12}R^{12}$ ,  $-NR^{21}C(O)R^{18}$ ,  
25  $OC(O)NR^{12}R^{12}$ ,  $-C(O)R^{19}$ ,  $-S(O)R^{18}$ ,  $-S(O)_2R^{18}$ ,  
 $-S(O)_2NR^{12}R^{12}$ ,  $-NO_2$ ,  $NR^{21}C(O)OR^{17}$ ,  
 $-C(NR^{12}R^{12})=NR^{12}$ , or  $-C(R^{13})=NOH$ ; or if  $Q^1$   
is  $C(O)OH$  and  $R^{22}$  is  $-OH$ ,  $-SH$ ,  $CHR^7OH$  or  
 $-NHR^3$ , then  $Q^1$  and  $R^{22}$  and the carbons  
30 through which they are attached may form a  
heterocyclic ring by loss of water;  
 $Q^2$  is  $OR^3$ ;

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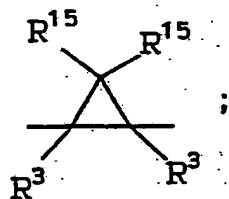
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W is O, S, or  $\text{NR}^3$ ;

$\text{X}^1$  is O, S,  $-\text{S}(\text{O})-$ ,  $-\text{S}(\text{O})_2-$ ,  $-\text{N}(\text{R}^3)-$ , or  $-\text{CR}^3\text{R}^3-$ ;

$\text{X}^2$  and  $\text{X}^3$  are independently O, S,  $\text{S}(\text{O})$ ,  $\text{S}(\text{O})_2$ ,  $\text{CR}^3\text{R}^{16}$ , or a bond;

Y is  $-\text{CR}^3=\text{CR}^3-$ ,  $-\text{C}=\text{C}-$ ,  $-\text{CR}^3\text{R}^3-\text{X}^1-$ ,  $-\text{X}^1-\text{CR}^3\text{R}^3-$ ,  $-\text{CR}^3\text{R}^3-\text{X}^1-\text{CR}^3\text{R}^3-$ ,  $-\text{C}(\text{O})-$ ,  $-\text{NR}^3\text{C}(\text{O})-$ ,  $-\text{C}(\text{O})\text{NR}^3-$ , O, S,  $\text{NR}^3$ , or



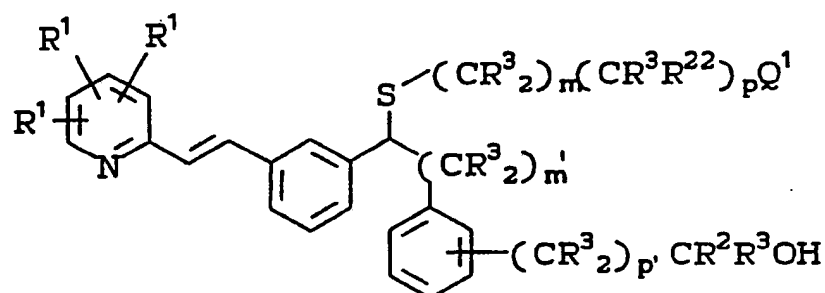
$\text{Z}^1$  and  $\text{Z}^2$  are independently  $-\text{HET}(-\text{R}^3-\text{R}^5)-$  or a bond;

HET is the diradical of a benzene, a pyridine, a furan, or a thiophene;

or a pharmaceutically acceptable salt thereof.

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2. A compound of Claim 1 of the Formula:



Ia

wherein:

$R^1$  is H, halogen, lower alkyl, polyhalo lower alkyl, lower alkoxy or adjacent  $R^1$ 's and the carbons through which they are attached may form a saturated ring of 5 to 7 carbon atoms;

$R^{22}$  is  $R^3$ ,  $-CH_2OR^3$ , or  $-CH_2SR^2$ ;

$Q^1$  is  $-C(O)OH$ , 1H(or 2H)-tetrazol-5-yl,  $-C(O)NHS(O)_2R^{13}$ ,  $-C(O)NR^{12}R^{12}$ , or  $-NHS(\bar{O})_2R^{13}$ ;

$m'$  is 2 or 3;

$p'$  is 0 or 1; and

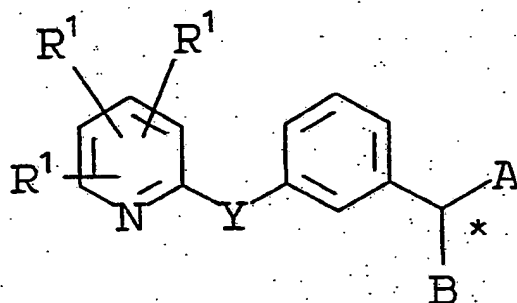
$m + p$  is 1-5;

or a pharmaceutically acceptable salt thereof.

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3. A compound of Claim 1 of the Formula:



1b

wherein the substituents are as follows:

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EX. #	R <sup>1</sup>	R <sup>1</sup>	Y	A	B	
1	R	5-Me	6-Me	CH=CH	(CH <sub>2</sub> ) <sub>2</sub> (1,2-phe)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>2</sub> CH <sub>2</sub> COOH
2	R	5-CF <sub>3</sub>	H	CH=CH	(CH <sub>2</sub> ) <sub>2</sub> (1,2-phe)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>2</sub> CH <sub>2</sub> COOH
3	R	6-Cl	H	CH=CH	(CH <sub>2</sub> ) <sub>2</sub> (1,2-phe)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>2</sub> CH <sub>2</sub> COOH
4	R	5-n-Bu	H	CH=CH	(CH <sub>2</sub> ) <sub>2</sub> (1,2-phe)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>2</sub> CH <sub>2</sub> COOH
5	R	4-Me	6-Me	CH=CH	(CH <sub>2</sub> ) <sub>2</sub> (1,2-phe)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>2</sub> CH <sub>2</sub> COOH
6	R	H	H	CH=CH	(CH <sub>2</sub> ) <sub>2</sub> (1,2-phe)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>2</sub> CH <sub>2</sub> COOH
7	R	5-OMe	H	CH=CH	(CH <sub>2</sub> ) <sub>2</sub> (1,2-phe)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>2</sub> CH <sub>2</sub> COOH
8	R	5,6-(CH <sub>2</sub> ) <sub>4</sub> -	H	CH=CH	(CH <sub>2</sub> ) <sub>2</sub> (1,2-phe)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>2</sub> CH <sub>2</sub> COOH
9	R	5,6-(CH <sub>2</sub> ) <sub>3</sub> -	H	CH=CH	(CH <sub>2</sub> ) <sub>2</sub> (1,2-phe)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>2</sub> CH <sub>2</sub> COOH
10	R	5-pH	H	CH=CH	(CH <sub>2</sub> ) <sub>2</sub> (1,2-phe)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>2</sub> CH <sub>2</sub> COOH
11	R	6-i-Pr	H	CH=CH	(CH <sub>2</sub> ) <sub>2</sub> (1,2-phe)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>2</sub> CH <sub>2</sub> COOH
12	R	5-Et	6-Me	CH=CH	(CH <sub>2</sub> ) <sub>2</sub> (1,2-phe)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>2</sub> CH <sub>2</sub> COOH
13	R	6-n-Bu	H	CH=CH	(CH <sub>2</sub> ) <sub>2</sub> (1,2-phe)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>2</sub> CH <sub>2</sub> COOH
14	R	6-c-Pen	H	CH=CH	(CH <sub>2</sub> ) <sub>2</sub> (1,2-phe)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>2</sub> CH <sub>2</sub> COOH
15	R	5-CF <sub>3</sub>	6-c-Bu	CH=CH	(CH <sub>2</sub> ) <sub>2</sub> (1,2-phe)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>2</sub> CH <sub>2</sub> COOH
16	R	5-F	6-i-Bu	CH=CH	(CH <sub>2</sub> ) <sub>2</sub> (1,2-phe)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>2</sub> CH <sub>2</sub> COOH
17	R	6-(4-F-Ph)	H	CH=CH	(CH <sub>2</sub> ) <sub>2</sub> (1,2-phe)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>2</sub> CH <sub>2</sub> COOH
18	R	5-CF <sub>3</sub> CH <sub>2</sub>	6-C-Pr-CH <sub>2</sub>	CH=CH	(CH <sub>2</sub> ) <sub>2</sub> (1,2-phe)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>2</sub> CH <sub>2</sub> COOH
19	R	5-Cl	6-Ac	CH=CH	(CH <sub>2</sub> ) <sub>2</sub> (1,2-phe)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> CH(CH <sub>3</sub> )CH <sub>2</sub> Tz

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EX. *	R <sup>1</sup>	R <sup>1</sup>	Y	A	B
20	RS	5,6-CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> CHMe-	CH=CH	(CH <sub>2</sub> ) <sub>2</sub> ((4-F-1,2-phe)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> CH(CH <sub>3</sub> )CH <sub>2</sub> Tz
21	R	5,6-(CH <sub>2</sub> ) <sub>6</sub> -	CH <sub>2</sub> O	(CH <sub>2</sub> ) <sub>2</sub> (4-Cl-1,2-phe)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>2</sub> CH <sub>2</sub> COOH
22	RS	6-(2-Thia)	H	(CH <sub>2</sub> ) <sub>2</sub> (3-CF <sub>3</sub> -1,2-phe)CHMeOH	SCH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> COOH
23	S	5-Et	6-Et	CH <sub>2</sub> (2,3-Thio)C(CH <sub>2</sub> ) <sub>2</sub> OH	SCH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> COOH
24	R	6-(2-Fu)	H	CH <sub>2</sub> (2,3-fur)C(CH <sub>2</sub> ) <sub>2</sub> OH	SCH <sub>2</sub> C(Me) <sub>2</sub> CONHSO <sub>2</sub> CF <sub>3</sub>
25	S	6-Th	H	(CH <sub>2</sub> ) <sub>2</sub> (1,2-phe)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>2</sub> CH <sub>2</sub> CO <sub>2</sub> H
26	R	5-CF <sub>3</sub> CH <sub>2</sub> CH <sub>2</sub>	6-EtO	(CH <sub>2</sub> ) <sub>2</sub> (4-Cl-1,2-phe)C(Me) <sub>2</sub> OH	OCH <sub>2</sub> C(CH <sub>3</sub> ) <sub>2</sub> CH <sub>2</sub> CO <sub>2</sub> H
27	RS	5-1-Pr	6-MeOCH <sub>2</sub>	(CH <sub>2</sub> ) <sub>2</sub> (4-Cl-1,2-phe)C(Me) <sub>2</sub> OH	OCH <sub>2</sub> CH(CH <sub>3</sub> )CH <sub>2</sub> Tz
28	RS	3-F	6-c-PrSCH <sub>2</sub>	(CH <sub>2</sub> ) <sub>2</sub> (4-F-1,2-phe)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> CH(C <sub>2</sub> H <sub>5</sub> )CH <sub>2</sub> CONME <sub>2</sub>
29	S	5-c-Pr	6-Me	CH <sub>2</sub> (4-F-1,2-phe)CMePhOH	SCH <sub>2</sub> CH <sub>2</sub> CO <sub>2</sub> H
30	R	5,6-(CH <sub>2</sub> ) <sub>4</sub> -		(CH <sub>2</sub> ) <sub>2</sub> (3-Cl-1,2-phe)CHMeOH	SCH <sub>2</sub> CH <sub>2</sub> CONHS(O) <sub>2</sub> Ph
31	R	5,6-CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> CH(Me)-		(CH <sub>2</sub> ) <sub>2</sub> (5-F-1,2-phe)CMeCF <sub>3</sub> OH	SCH <sub>2</sub> CH <sub>2</sub> CONHS(O) <sub>2</sub> CH <sub>3</sub>
32	RS	5-MeOCH <sub>2</sub>	6-EtSCH <sub>2</sub>	(6-CF <sub>3</sub> -1,2-phe)CHCF <sub>3</sub> OH	SCH <sub>2</sub> CH <sub>2</sub> CONHS(O) <sub>2</sub> CF <sub>3</sub>
33	R	5-Me	6-C <sub>3</sub> H <sub>5</sub>	(CH <sub>2</sub> ) <sub>2</sub> (4-CF <sub>3</sub> -1,2-phe)C(CF <sub>3</sub> ) <sub>2</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>2</sub> CH <sub>2</sub> CONHS(O) <sub>2</sub> Ph
34	R	5-1-C <sub>3</sub> H <sub>5</sub>	6-Me	(CH <sub>2</sub> ) <sub>2</sub> (4-F-1,3-phe)CMeEtOH	SCH <sub>2</sub> CH <sub>2</sub> C(CH <sub>3</sub> ) <sub>2</sub> CO <sub>2</sub> H
35	RS	6-(2-Py)	H	(CH <sub>2</sub> ) <sub>2</sub> (4-F-1,4-phe)C(CH <sub>2</sub> ) <sub>2</sub> OH	SCH <sub>2</sub> CH(C <sub>2</sub> H <sub>5</sub> )Tz
36	S	5-Cl	6-OF <sub>3</sub> CH <sub>2</sub> CHMe	(CH <sub>2</sub> ) <sub>2</sub> (4-F-1,2-phe)C(CH <sub>2</sub> ) <sub>3</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>2</sub> NHS(O) <sub>2</sub> CF <sub>3</sub>
37	R	4-F	5,6-(CH <sub>2</sub> ) <sub>4</sub>	(CH <sub>2</sub> ) <sub>2</sub> (4-F-1,2-phe)C(CH <sub>2</sub> ) <sub>4</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>3</sub> CH <sub>2</sub> CO <sub>2</sub> H
38	R	3-MeO	5-(CF <sub>3</sub> ) <sub>2</sub> CHCH <sub>2</sub>	(CH <sub>2</sub> ) <sub>2</sub> (4-F-1,2-phe)C(CH <sub>2</sub> ) <sub>5</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>4</sub> CH <sub>2</sub> CO <sub>2</sub> H
39	R	5-F	6-(2-Py)	(CH <sub>2</sub> ) <sub>2</sub> (2,5-fur)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> C(CH <sub>2</sub> ) <sub>5</sub> CH <sub>2</sub> CO <sub>2</sub> H
40	RS	6-(2-Ox)	H	(CH <sub>2</sub> ) <sub>2</sub> (1,2-phe)C(Me) <sub>2</sub> OH	SCH <sub>2</sub> C(CH <sub>3</sub> ) <sub>2</sub> CH <sub>2</sub> CO <sub>2</sub> H

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4. A pharmaceutical composition comprising a therapeutically effective amount of a compound of Claim 1 and a pharmaceutically acceptable carrier.

5                   5. The pharmaceutical composition of Claim 4 additionally comprising an effective amount of a second active ingredients selected from the group consisting of non-steroidal anti-inflammatory drugs; peripheral analgesic agents; cyclooxygenase  
10 inhibitors; leukotriene antagonists; leukotriene biosynthesis inhibitors; H<sub>1</sub>- or H<sub>2</sub>-receptor antagonists; antihistaminic agents; prostaglandin antagonists; and ACE antagonists.

15                   6. A pharmaceutical composition of Claim 5, wherein the second active ingredient is a non-steroidal anti-inflammatory drug.

20                   7. A pharmaceutical composition of Claim 6, wherein the weight ratio of said compound of Claim 1 to said second active ingredient ranges from about 1000:1 to 1:1000.

25                   8. A method of preventing the synthesis, the action, or the release of SRS-A or leukotrienes in a mammal which comprises administering to said mammal an effective amount of a compound of Claim 1.

30                   9. The method of Claim 9 wherein the mammal is man.

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10. A method of treating asthma in a mammal comprising administering to a mammal in need of such treatment a therapeutically effective amount of a compound of Claim 1.

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11. The method of Claim 10 wherein the mammal is man.

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12. A method of treating inflammatory diseases of the eye in mammal which comprises administering to a mammal in need of such treatment a therapeutically effective amount of a compound of Claim 1.

15

13. A method of formula (I), or a pharmaceutically acceptable salt, thereof, as defined in claim 1, 2, or 3, for use in preventing the synthesis, the action, or the release of SRS-A or leukotrienes in a mammal.

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14. Use of a compound of formula (I), or a pharmaceutically acceptable salt thereof, as defined in claim 1, 2, or 3, in the manufacture of a medicament for treating asthma or inflammatory diseases of the eye.

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15. A leukotriene antagonist pharmaceutical composition comprising an acceptable leukotriene antagonistic amount of a compound of formula (I), or a pharmaceutically acceptable salt thereof, as defined in claim 1, 2, or 3, in association with a pharmaceutically acceptable carrier.

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## INTERNATIONAL SEARCH REPORT

PCT/CA 93/00145

International Application No

**I. CLASSIFICATION OF SUBJECT MATTER** (If several classification symbols apply, indicate all)<sup>6</sup>

According to International Patent Classification (IPC) or to both National Classification and IPC

Int.Cl. 5 C07D213/65;	C07D213/32;	C07D213/61;	C07D215/12
C07D401/12;	C07D417/04;	A61K31/44;	A61K31/41

**II. FIELDS SEARCHED**Minimum Documentation Searched<sup>7</sup>

Classification System

Classification Symbols

Int.Cl. 5

C07D ; A61K

Documentation Searched other than Minimum Documentation  
to the Extent that such Documents are Included in the Fields Searched<sup>8</sup>**III. DOCUMENTS CONSIDERED TO BE RELEVANT<sup>9</sup>**

Category <sup>10</sup>	Citation of Document, <sup>11</sup> with indication, where appropriate, of the relevant passages <sup>12</sup>	Relevant to Claim No. <sup>13</sup>
Y	US,A,5 004 743 (MERCK FROSST CANADA, INC) 2 April 1991 cited in the application *see especially definitions of Q1 and Q2* ---	1-15
Y	EP,A,0 318 093 (MERCK FROSST CANADA, INC) 31 May 1989 cited in the application *see definitions of Q1 and Q2* ---	1-15
Y	US,A,4 918 081 (RORER PHARMACEUTICAL CORP.) 17 April 1990 cited in the application *see especially definitions of R,Z and Y* ---	1-15
P,Y	EP,A,0 480 708 (MERCK FROSST CANADA INC.) 15 April 1992 *see definitions of Q1 and Q2* --- -/-	1-15

<sup>10</sup> Special categories of cited documents:<sup>"A"</sup> document defining the general state of the art which is not considered to be of particular relevance<sup>"E"</sup> earlier document but published on or after the international filing date<sup>"L"</sup> document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)<sup>"O"</sup> document referring to an oral disclosure, use, exhibition or other means<sup>"P"</sup> document published prior to the international filing date but later than the priority date claimed<sup>"T"</sup> later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention<sup>"X"</sup> document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step<sup>"Y"</sup> document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.<sup>"A"</sup> document member of the same patent family**IV. CERTIFICATION**

Date of the Actual Completion of the International Search

06 JULY 1993

Date of Mailing of this International Search Report

19. 07. 93

International Searching Authority

EUROPEAN PATENT OFFICE

Signature of Authorized Officer

SCRUTON-EVANS I.

**Category <sup>B</sup>**

Citation of Document, with indication, where appropriate, of the relevant passages

Relevant to Claim No.

Category	Citation or Document, with location, where appropriate, of the relevant passage	
A	<p>EP,A,O 110 405 (USV PHARMACEUTICAL CORPORATION) 13 June 1984</p> <p style="text-align: center;">---</p>	1-15
Y	<p>EP,A,O 181 568 (USV PHARMACEUTICAL CORPORATION) 21 May 1986 *see definitions of R, and page 44, lines 1-6*</p> <p style="text-align: center;">-----</p>	1-15

# ANNEX TO THE INTERNATIONAL SEARCH REPORT ON INTERNATIONAL PATENT APPLICATION NO.

CA 9300145  
SA 72368

This annex lists the patent family members relating to the patent documents cited in the above-mentioned international search report.  
The members are as contained in the European Patent Office EDP file on  
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06/07/93

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